



# La Química, la catálisis y la vida



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**H** ≡ Hidrógeno  
**C** ≡ Carbono  
**N** ≡ Nitrógeno  
**O** ≡ Oxígeno  
**Ca** ≡ Calcio  
**Si** ≡ Silicio  
**Fe** ≡ Hierro

The collage features several chemical structures: a carbon dioxide molecule (CO<sub>2</sub>) with a central carbon atom (C) and two oxygen atoms (O); a calcium carbonate molecule (CaCO<sub>3</sub>) with a central carbon atom (C) and three oxygen atoms (O) and a calcium atom (Ca); a silicon dioxide molecule (SiO<sub>2</sub>) with a central silicon atom (Si) and two oxygen atoms (O); a water molecule (H<sub>2</sub>O) with one oxygen atom (O) and two hydrogen atoms (H); and a DNA double helix structure. A red blood cell is shown with the label 'Fe'.

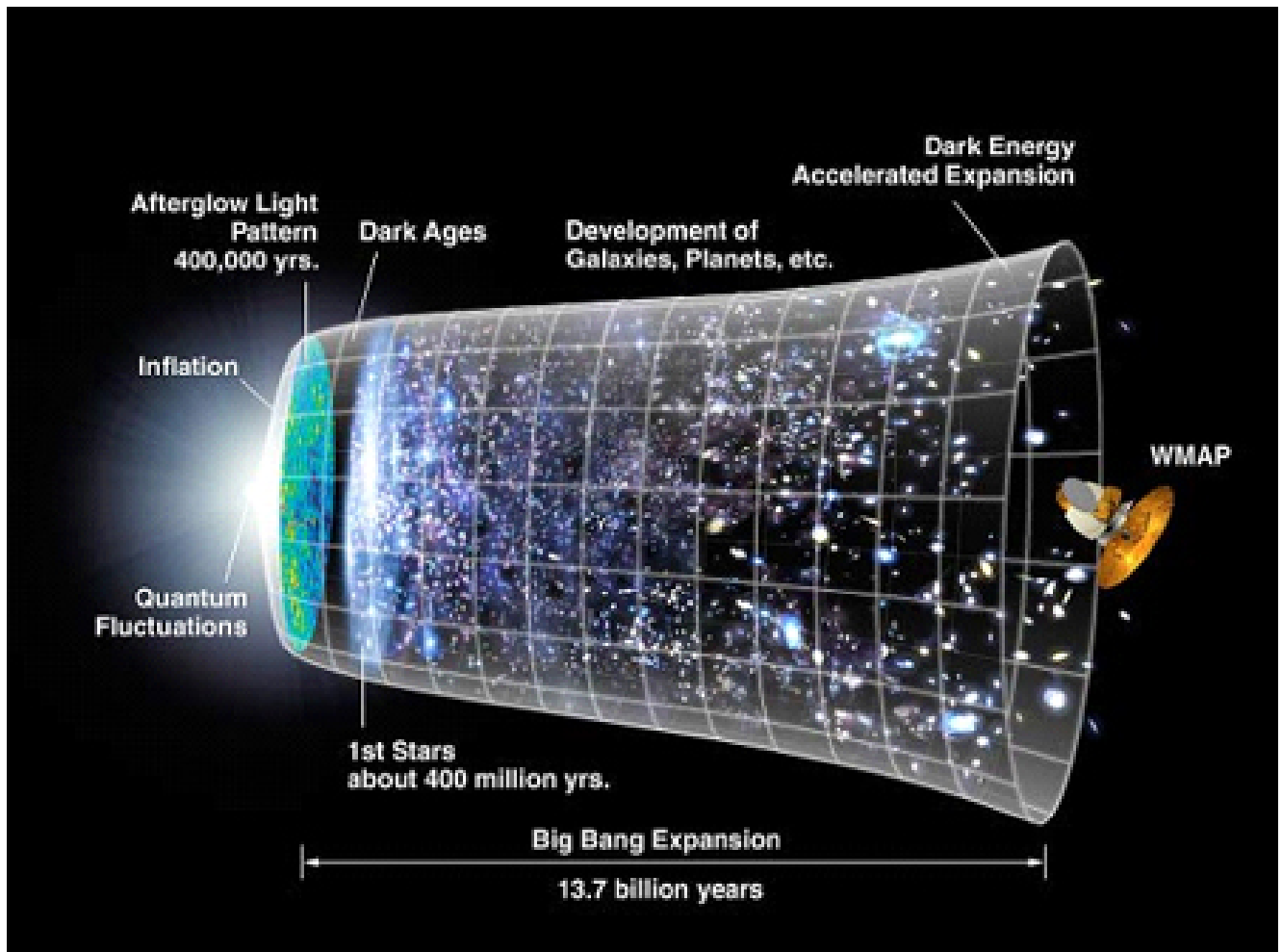
1 1.008 H																	2 4.0026 He
3 6.941 Li	4 9.0122 Be											5 10.81 B	6 12.01 C	7 14.007 N	8 15.999 O	9 18.998 F	10 20.179 Ne
11 22.990 Na	12 24.305 Mg											13 26.982 Al	14 28.086 Si	15 30.974 P	16 32.06 S	17 35.453 Cl	18 39.948 Ar
19 39.098 K	20 40.078 Ca	21 44.956 Sc	22 47.867 Ti	23 50.942 V	24 51.996 Cr	25 54.938 Mn	26 55.845 Fe	27 58.933 Co	28 58.933 Ni	29 63.546 Cu	30 65.38 Zn	31 69.723 Ga	32 72.59 Ge	33 74.922 As	34 78.96 Se	35 79.904 Br	36 83.80 Kr
37 85.468 Rb	38 87.62 Sr	39 88.906 Y	40 91.224 Zr	41 92.905 Nb	42 95.94 Mo	43 95.94 Tc	44 101.07 Ru	45 102.906 Rh	46 106.42 Pd	47 107.868 Ag	48 112.41 Cd	49 114.82 In	50 118.71 Sn	51 121.76 Sb	52 127.60 Te	53 126.905 I	54 131.29 Xe
55 132.905 Cs	56 137.33 Ba	57 138.905 La	58 178.49 Hf	59 180.948 Ta	60 183.85 W	61 186.207 Re	62 186.207 Os	63 192.22 Ir	64 195.08 Pt	65 196.967 Au	66 200.59 Hg	67 204.383 Tl	68 207.2 Pb	69 208.980 Bi	70 209 Po	71 209 At	72 222 Rn
87 (223) Fr	88 226.025 Ra	89 227.028 Ac															

58 140.12 Ce	59 140.908 Pr	60 144.24 Nd	61 144.913 Pm	62 150.36 Sm	63 151.96 Eu	64 157.25 Gd	65 158.925 Tb	66 162.50 Dy	67 164.930 Ho	68 167.26 Er	69 168.934 Tm	70 173.054 Yb	71 174.967 Lu
90 232.038 Th	91 231.036 Pa	92 238.029 U	93 237.048 Np	94 244 Pu	95 244 Am	96 244 Cm	97 247 Bk	98 251 Cf	99 254 Es	100 257 Fm	101 257 Md	102 259 No	103 262 Lr

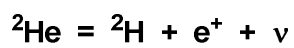
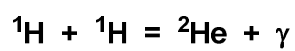
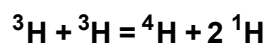
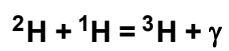
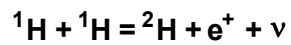
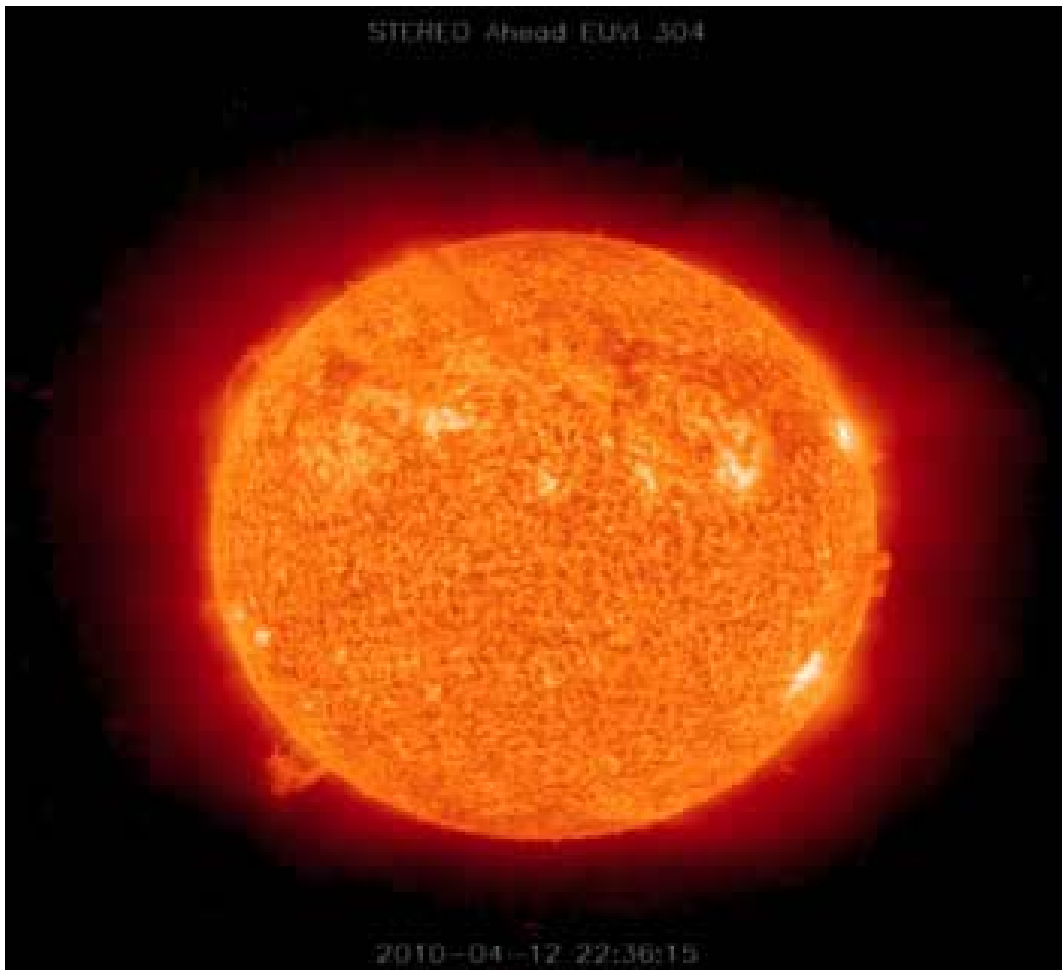
1 1.008 H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac															

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
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Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
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<b>H</b>																	<b>He</b>		
<b>Li</b>	<b>Be</b>											<b>B</b>	<b>C</b>	<b>N</b>	<b>O</b>	<b>F</b>	<b>Ne</b>		
<b>Na</b>	<b>Mg</b>													<b>Al</b>	<b>Si</b>	<b>P</b>	<b>S</b>	<b>Cl</b>	<b>Ar</b>
<b>K</b>	<b>Ca</b>	<b>Sc</b>	<b>Ti</b>	<b>V</b>	<b>Cr</b>	<b>Mn</b>	<b>Fe</b>	<b>Co</b>	<b>Ni</b>	<b>Cu</b>	<b>Zn</b>	<b>Ga</b>	<b>Ge</b>	<b>As</b>	<b>Se</b>	<b>Br</b>	<b>Kr</b>		
<b>Rb</b>	<b>Sr</b>	<b>Y</b>	<b>Zr</b>	<b>Nb</b>	<b>Mo</b>	<b>Tc</b>	<b>Ru</b>	<b>Rh</b>	<b>Pd</b>	<b>Ag</b>	<b>Cd</b>	<b>In</b>	<b>Sn</b>	<b>Sb</b>	<b>Te</b>	<b>I</b>	<b>Xe</b>		
<b>Ce</b>	<b>Ba</b>	<b>La</b>	<b>Hf</b>	<b>Ta</b>	<b>W</b>	<b>Re</b>	<b>Os</b>	<b>Ir</b>	<b>Pt</b>	<b>Au</b>	<b>Hg</b>	<b>Tl</b>	<b>Pb</b>	<b>Bi</b>	<b>Po</b>	<b>At</b>	<b>Rn</b>		
<b>Pr</b>	<b>Ra</b>	<b>Ac</b>																	
<b>Ce</b>	<b>Pr</b>	<b>Nd</b>	<b>Pm</b>	<b>Sm</b>	<b>Eu</b>	<b>Gd</b>	<b>Tb</b>	<b>Dy</b>	<b>Ho</b>	<b>Er</b>	<b>Tm</b>	<b>Yb</b>	<b>Lu</b>						
<b>Th</b>	<b>Pa</b>	<b>U</b>	<b>Np</b>	<b>Pu</b>	<b>Am</b>	<b>Cm</b>	<b>Bk</b>	<b>Cf</b>	<b>Es</b>	<b>Fm</b>	<b>Md</b>	<b>No</b>	<b>Lr</b>						

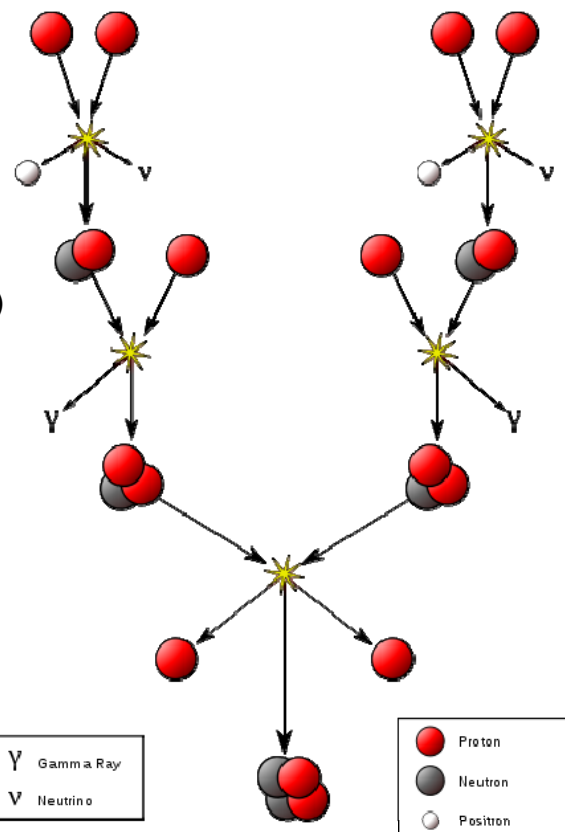


(1)

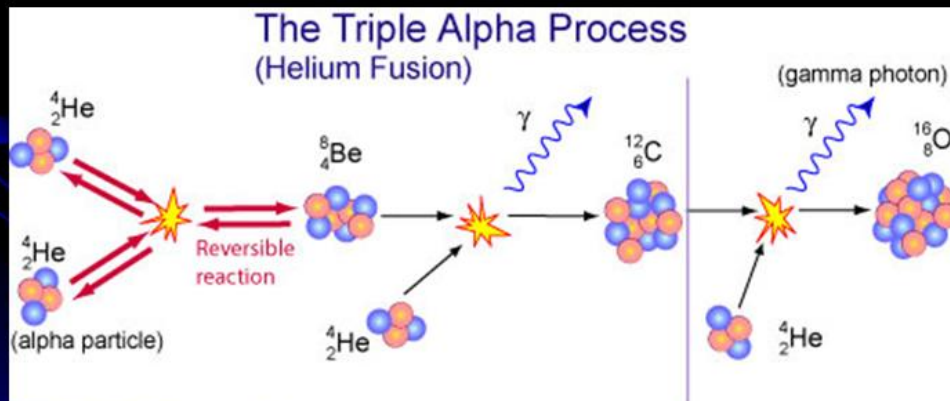
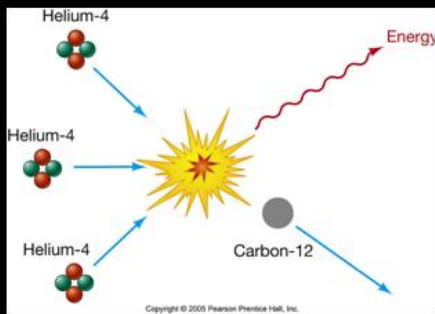
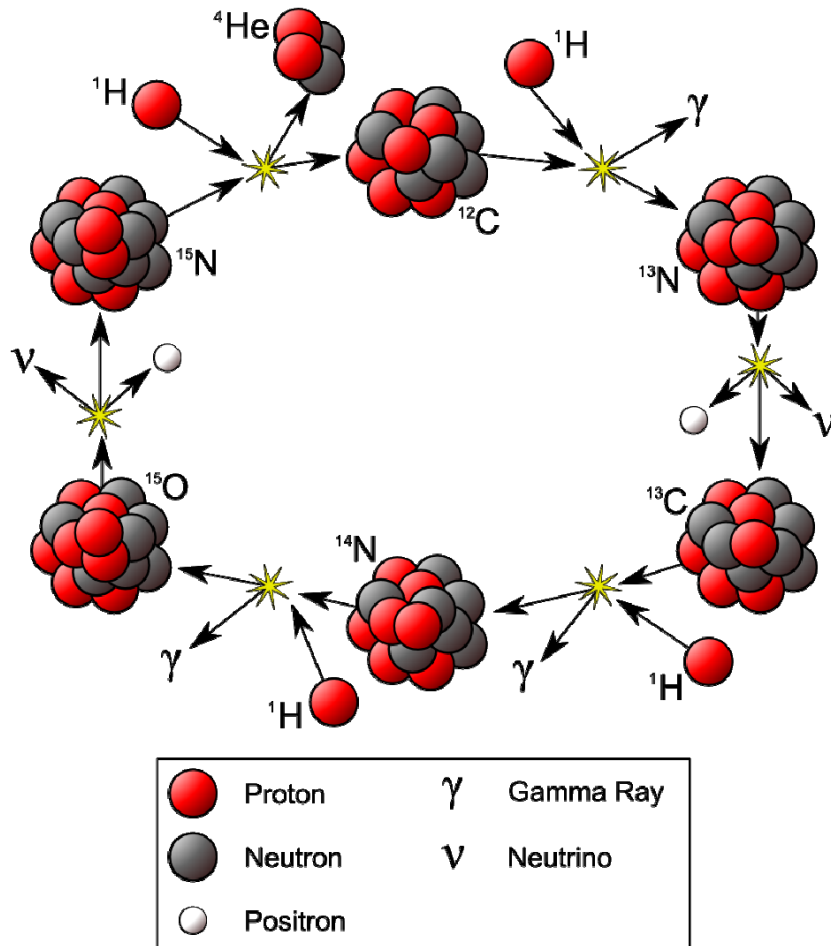
(2)

(3)

(26,72 MeV)



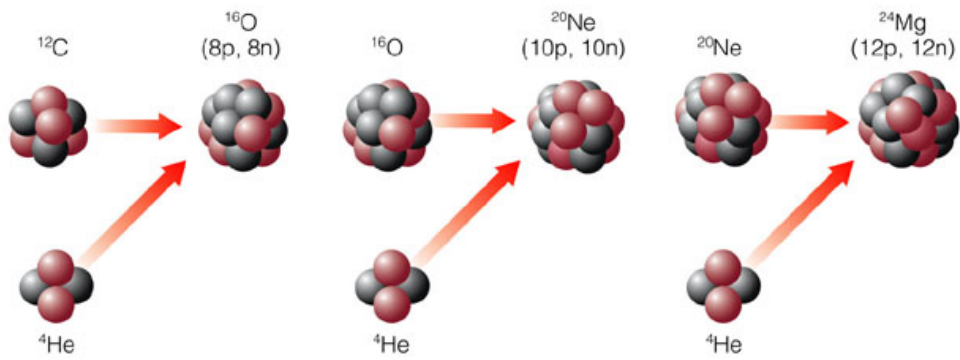




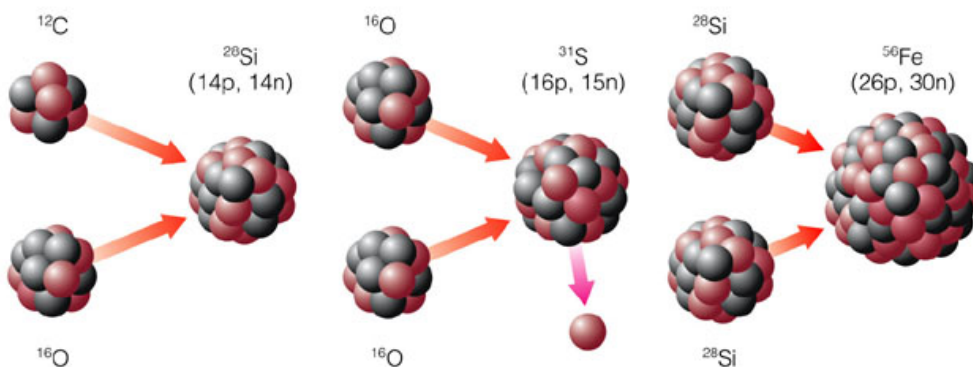


WE ARE  
MADE OF  
STAR STUFF  
C SAGAN

**Helium-capture reactions**

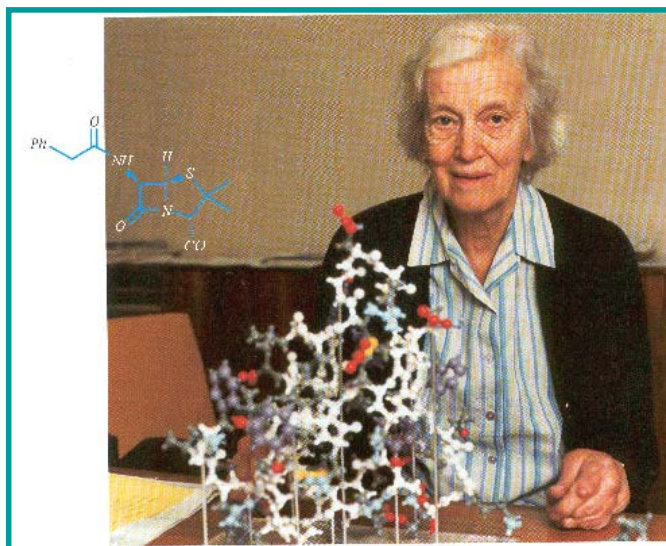
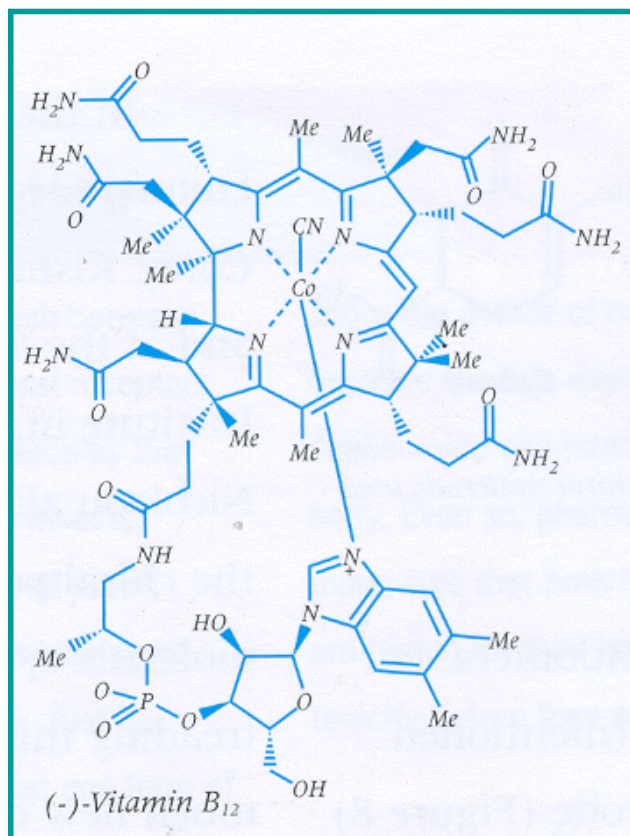


**Other reactions**



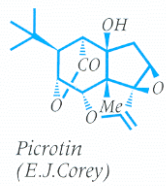
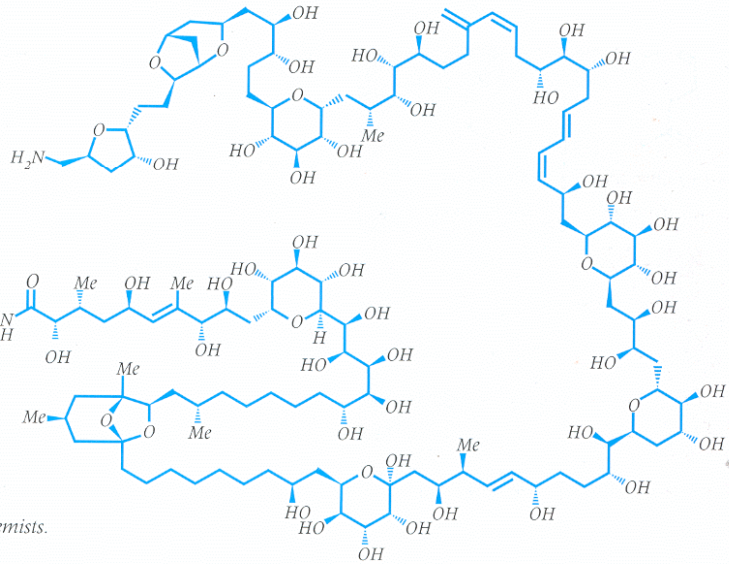
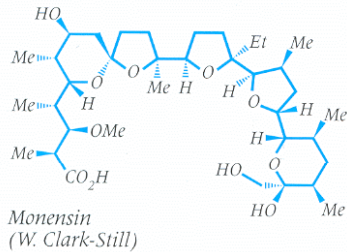
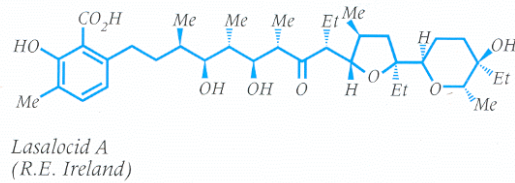
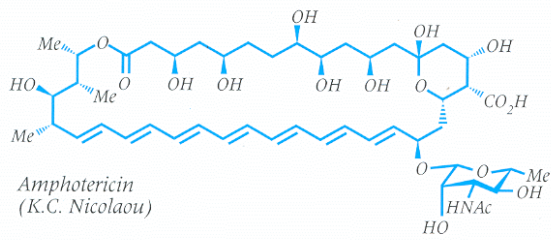
*« Before the word for it (in any language) came to be, there was chemistry. For one defining aspect of human beings has always been the meld of mind and hands in transforming matter. »*

Roald Hoffmann  
Premio Nobel de Química 1981.  
The New Chemistry. N. Hall, Ed.  
Cambridge Univ. Press, 2000

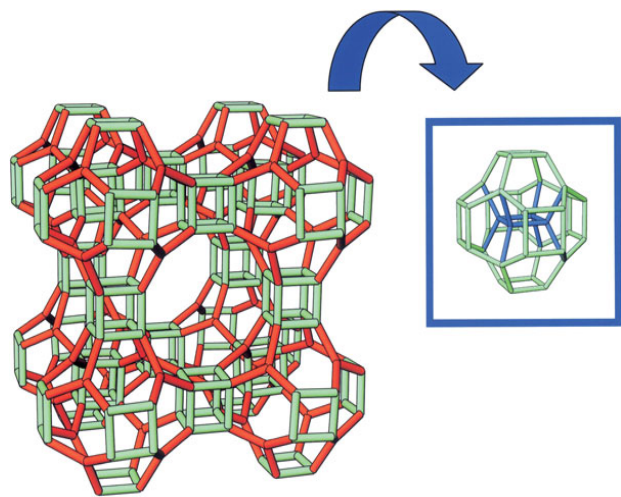
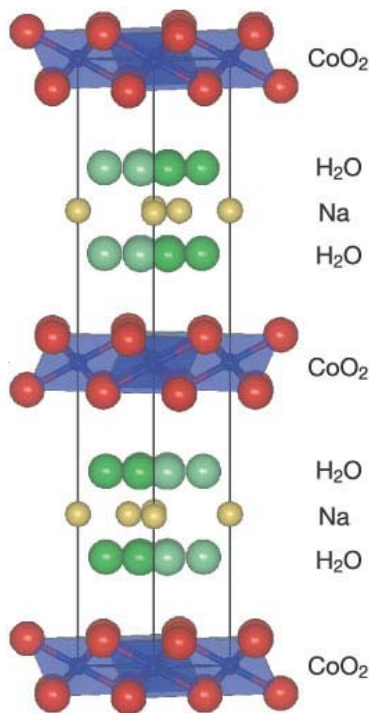


D. Hodgkin  
Premio Nobel, 1964





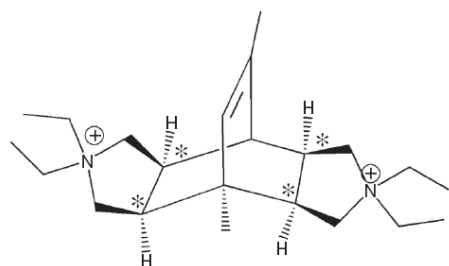
Some complex natural molecules synthesised by chemists.



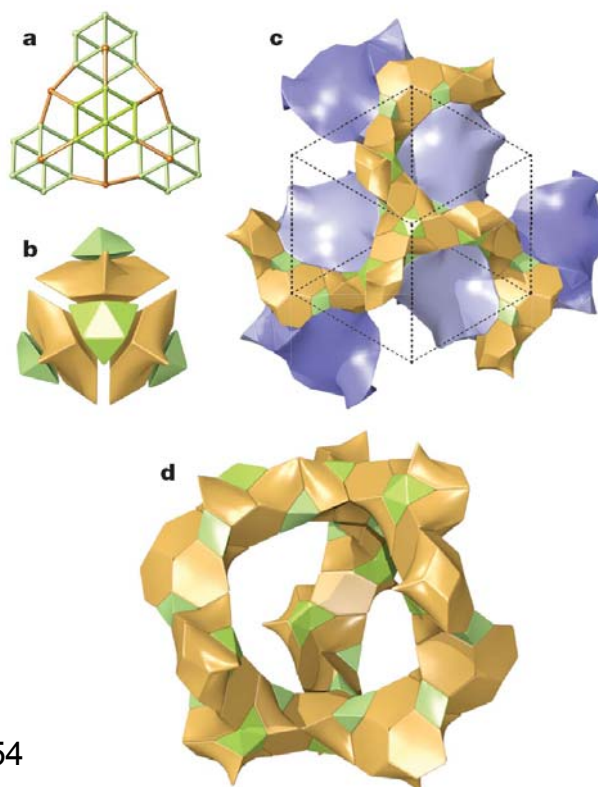
**Nature, 2002, 418, 515**

**Nature, 2003, 422, 53**

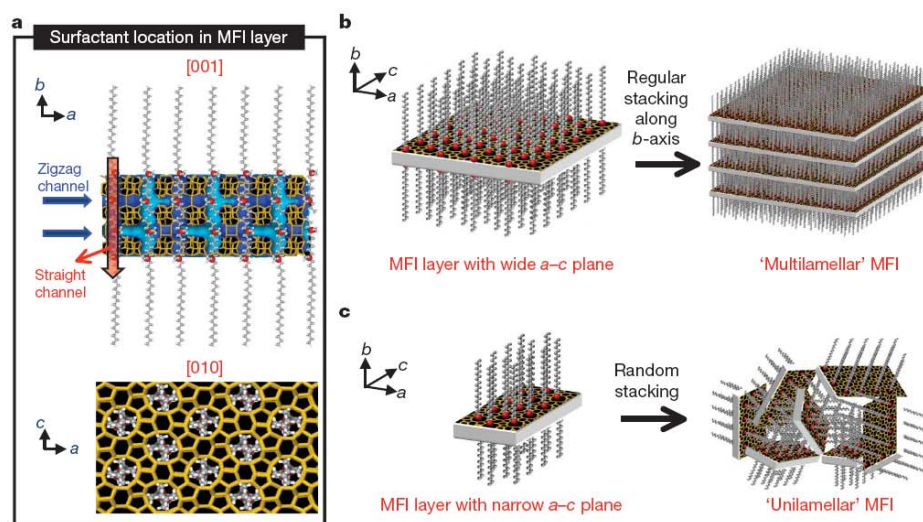




**Figure 1** | Structure of SDA2 used for synthesizing the ITQ-37 zeolite. SDA2 contains four chiral centres (marked with asterisks) in a meso conformation, making the overall molecule achiral.



A. Corma, *et al.* *Nature* **2009**, 458, 1154



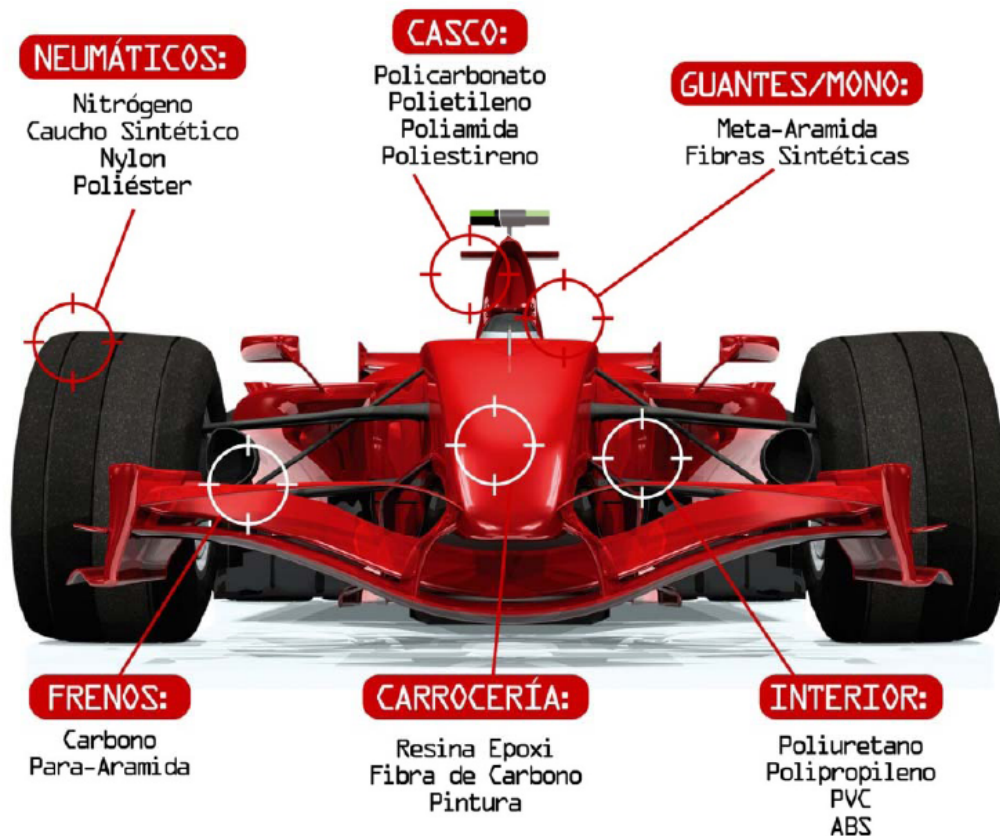
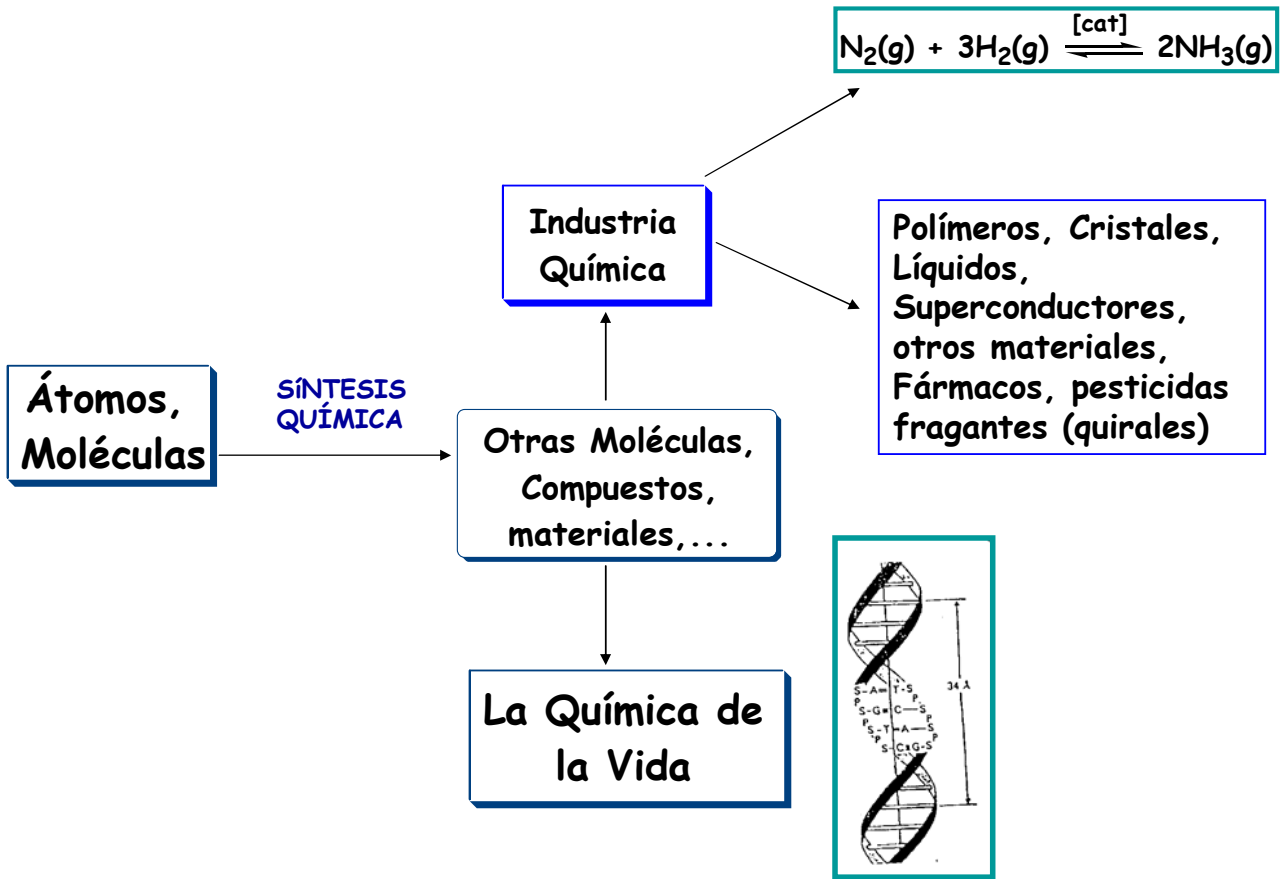
**Figure 2** | Crystallization of MFI nanosheets. **a**, Proposed structure model for the single MFI nanosheet. Surfactant molecules are aligned along the straight channel of MFI framework. Two quaternary ammonium groups (indicated as a red sphere) are located at the channel intersections; one is

inside the framework, and the other is at the pore mouth of the external surface. Many MFI nanosheets form either multilamellar stacking along the *b*-axis (**b**), or a random assembly of unilamellar structure (**c**).

**Table 1** | Catalytic conversion of bulky molecules over MFI zeolites

Reactions	Conventional MFI (Si/Al = 41)	Multilamellar MFI nanosheets (Si/Al = 48)	Unilamellar MFI nanosheets (Si/Al = 53)
Cracking of branched polyethylene (HDPE)	27	45	85
 Benzaldehyde + Resorcinol → Flavanone + Chalcone	16 (50/50/0)*	48 (62/28/10)*	76 (64/31/5)*
 Benzaldehyde + 1,2-cyclohexanediol → Diacetal	42	86	86

Catalytic activities were compared on the basis of the same weight of catalyst (see Methods for reaction conditions). \*The numbers in parentheses indicate percentage selectivity; (flavanone/chalcone/others). All other numbers indicate the percentage reactant conversion, reproducible within 3% over three runs. HDPE, high-density polyethylene.



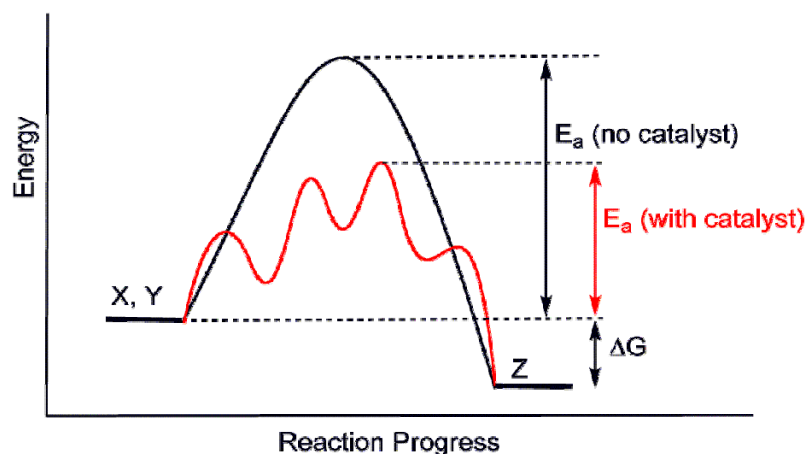
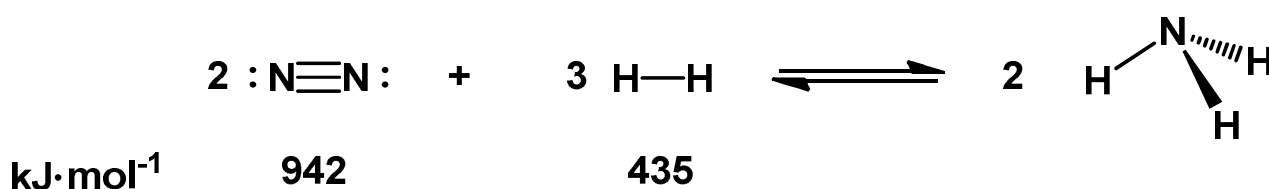
***“A catalyst is a substance which affects the rate of a chemical reaction without being part of its end products”***

W. Ostwald, 1900. Premio Nobel de Química, 1909.

***“A catalyst is a substance which increases the rate at which a chemical reaction approaches equilibrium without becoming itself permanently involved”***

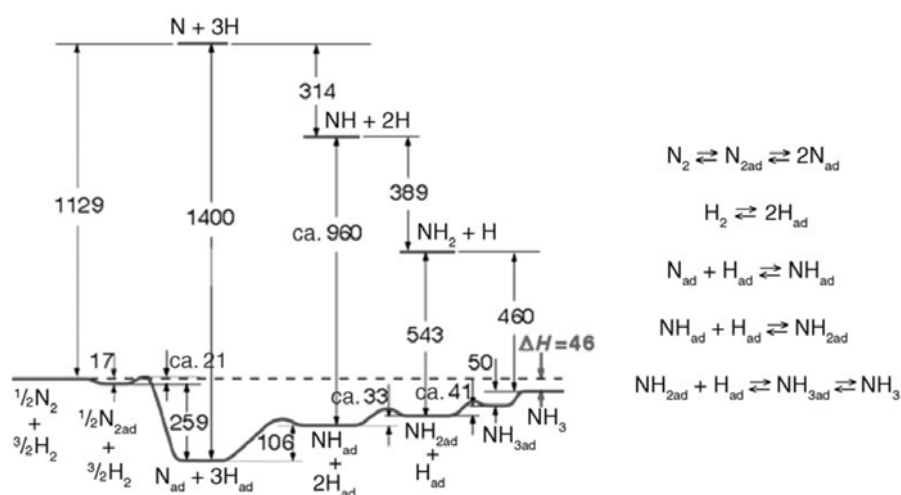
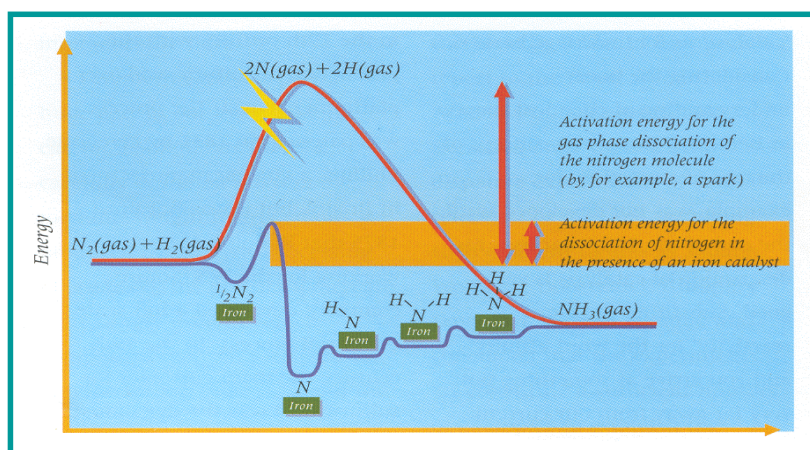
P. W. N. M. van Leewen, Homogeneous Catalysis

Kluwer Academic Publishers, Dordrecht, 2004.

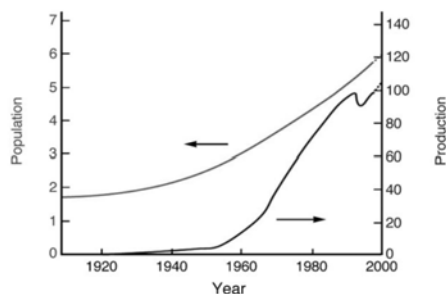


W. Crooks, 1898 (68th meeting of the British Association for the Advancement of Science):

*“... all civilized nations stand in deadly peril of not having enough to eat... the fixation of atmospheric nitrogen is one of the great discoveries awaiting the ingenuity of chemists.”*







Industry	Use
Fertilizer	Production of: ammonium sulfate, $(\text{NH}_4)_2\text{SO}_4$ ; ammonium phosphate, $(\text{NH}_4)_3\text{PO}_4$ ; ammonium nitrate, $\text{NH}_4\text{NO}_3$ ; urea, $(\text{NH}_2)_2\text{CO}$
Chemicals	Synthesis of: nitric acid, $\text{HNO}_3$ which is used in making explosives such as TNT (2,4,6-trinitrotoluene), nitroglycerine which is also used as a vasodilator (a substance that dilates blood vessels) and PETN (pentaerythritol nitrate); sodium hydrogen carbonate (sodium bicarbonate), $\text{NaHCO}_3$ ; sodium carbonate, $\text{Na}_2\text{CO}_3$ ; hydrogen cyanide (hydrocyanic acid), $\text{HCN}$ ; hydrazine, $\text{N}_2\text{H}_4$ (used in rocket propulsion systems).
Explosives	Ammonium nitrate, $\text{NH}_4\text{NO}_3$
Fibres & plastics	nylon, $-\text{[(CH}_2)_4\text{-CO-NH-(CH}_2)_6\text{-NH-CO]}-$ , and other polyamides.
Refrigeration	Used for making ice, large scale refrigeration plants, air-conditioning units in buildings and plants.
Pharmaceuticals	Used in the manufacture of drugs such as sulfonamide which inhibit the growth and multiplication of bacteria that require <i>p</i> -aminobenzoic acid (PABA) for the biosynthesis of folic acids; anti-malarials and vitamins such as the B vitamins nicotinamide (niacinamide) and thiamine.
Pulp & paper	Ammonium hydrogen sulfite, $\text{NH}_4\text{HSO}_3$ , enables some hardwoods to be used
Mining & metallurgy	Used in nitriding (bright annealing) steel; used in zinc and nickel extraction.



***"for his studies of chemical processes on solid surfaces".***

**G. Ertl, Premio Nobel de Química 2007**



***"in recognition of their contributions to the invention and development of chemical high pressure methods"***

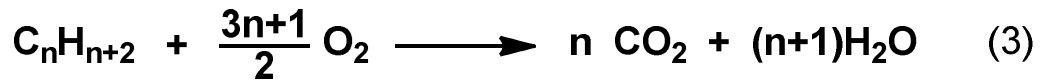
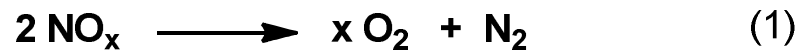
**C. Bosch, Premio Nobel de Química 1931 (F. Bergius)**



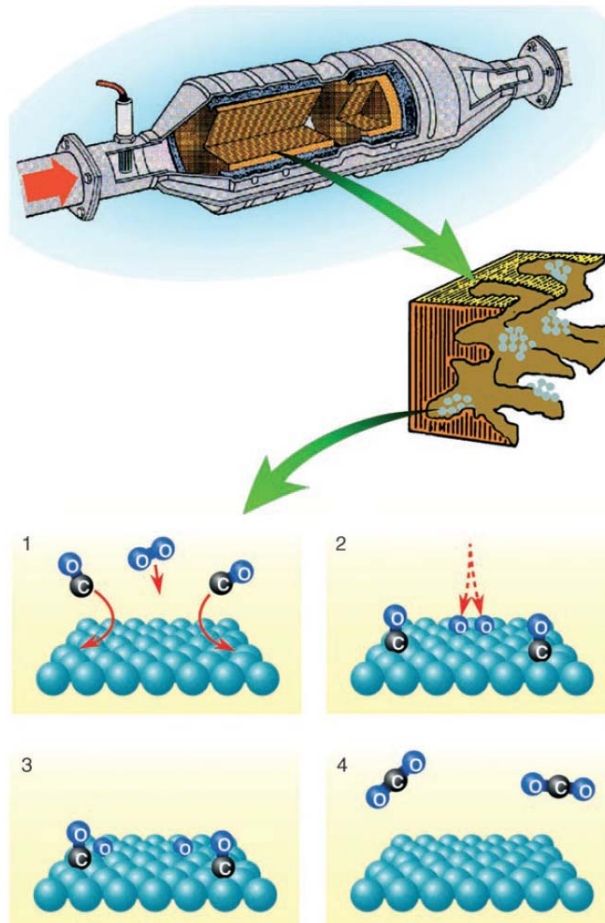
***"for the synthesis of ammonia from its elements"***

**F. Haber, Premio Nobel de Química 1918**

## Catalizadores de tres vías

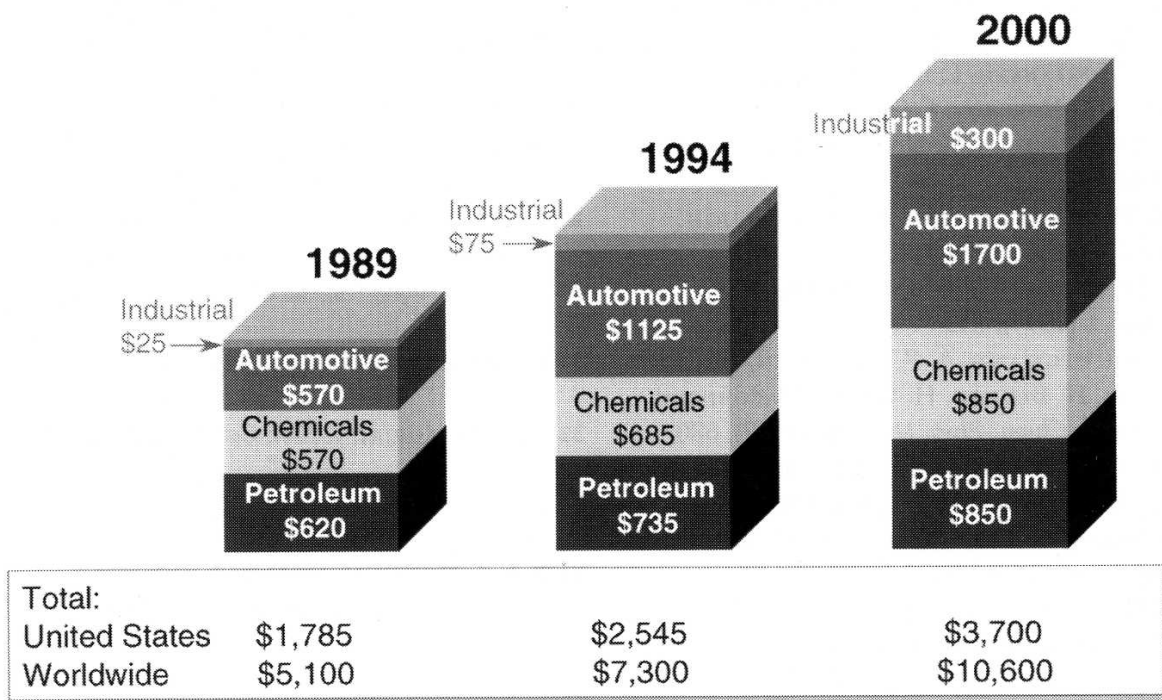


otras reacciones como:



## Development of the catalyst market

(in millions of dollars)



**Fig. 4.13** Diagram showing the development of the catalyst market. [40]

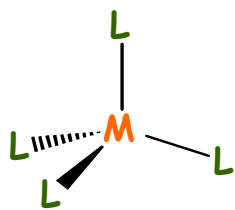


$\text{CoCl}_3(\text{NH}_3)_6$  (amarillo)

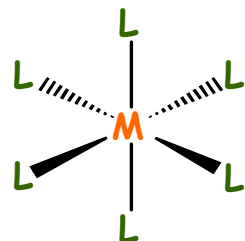
$\text{CoCl}_3(\text{NH}_3)_5$  (púrpura)

$\text{CoCl}_3(\text{NH}_3)_4$  (verde y violeta)

$\text{CoCl}_3(\text{NH}_3)_3$  (dos isómeros)



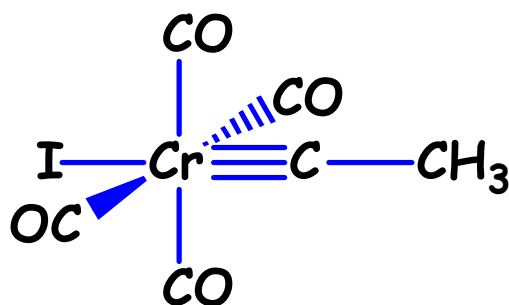
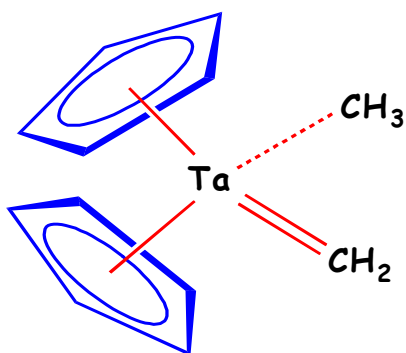
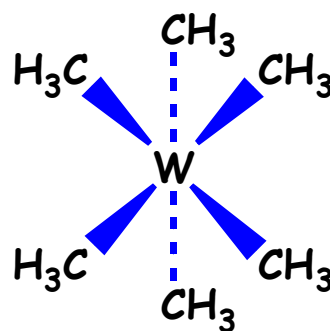
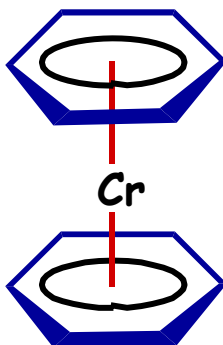
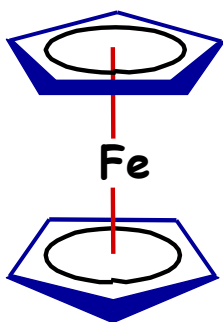
$\text{ML}_4$  (Td)



$\text{ML}_6$  (Oh)

A. Werner.

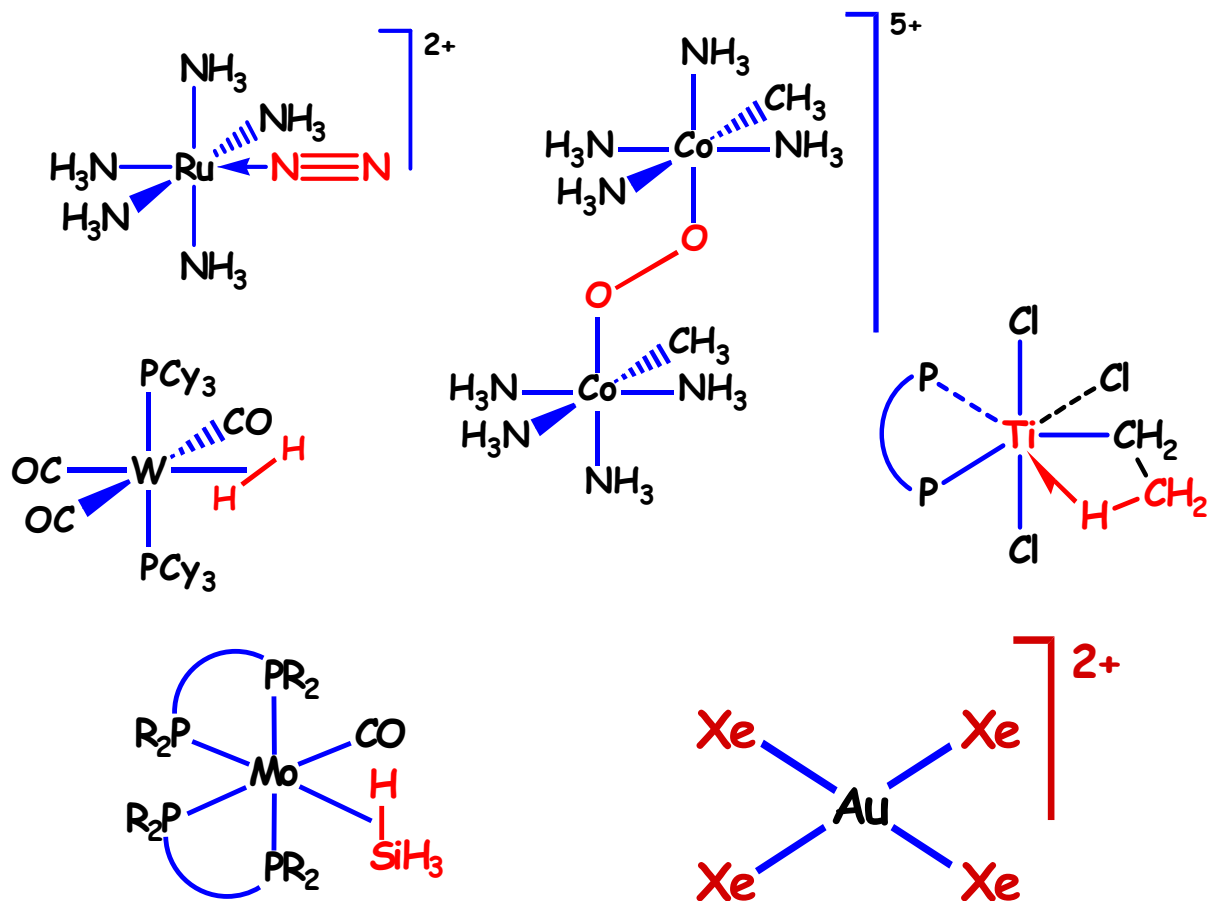
Premio Nobel de Química, 1913



E.O. Fischer, G. Wilkinson.

Premio Nobel de Química, 1973

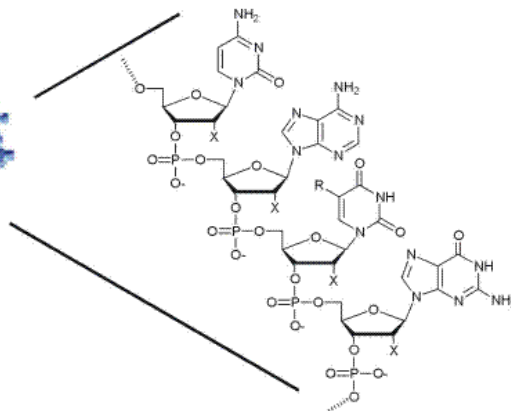
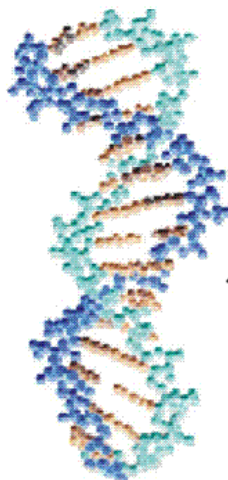
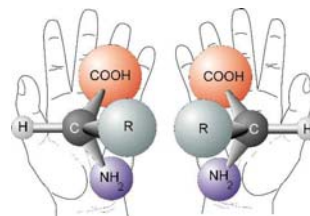




## Algunos procesos catalíticos homogéneos

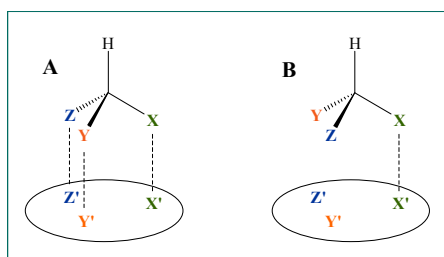
- ▶ Catálisis Ziegler – Natta para la producción de poliolefinas (metallocenos como catalizadores).
- ▶ Hidroformilación de alquenos catalizada mediante complejos de cobalto ( $\text{CoH}(\text{CO})_4$ ), o de rodio ( $\text{RhH}(\text{CO})(\text{PPh}_3)_3$ ). Síntesis de diversos productos oxigenados (alcoholes detergentes....).
- ▶ Síntesis del ácido acético: Monsanto (Rh) y Cativa (Ir)
- ▶ Metátesis de olefinas: carbenos de Schrock (Mo, W) y de Grubbs (Ru). **Premio Nobel de Química de 2005** (Y. Chauvin, R.H. Grubbs, R.R. Schrock).

**La Quiralidad es un fenómeno común en la naturaleza:**  
***Proteínas, Carbohidratos, Acidos Nucleicos***

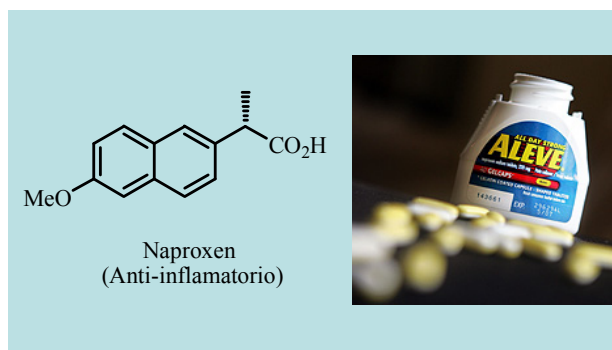
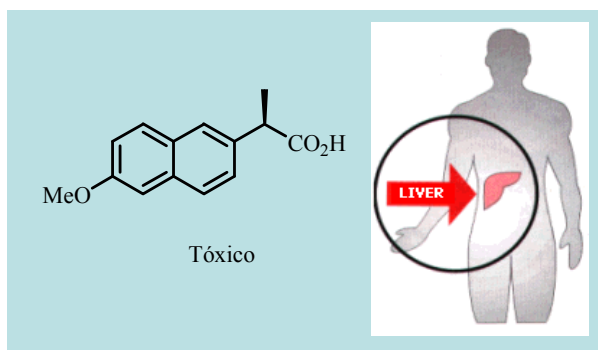


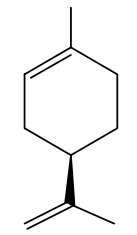
Las moléculas, como los objetos macroscópicos, tienen una estructura tridimensional. La quiralidad de una molécula es una consecuencia de su estructura espacial.

## ***Productos Quirales: Efecto Biológico***

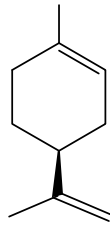


## **Enantiómeros y Propiedades Biológicas ⇨ *Naproxeno***



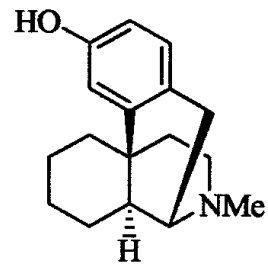


(R)-limonene

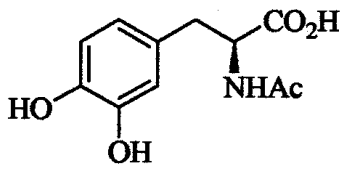


(S)-limonene

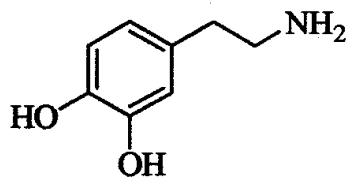
mirror plane



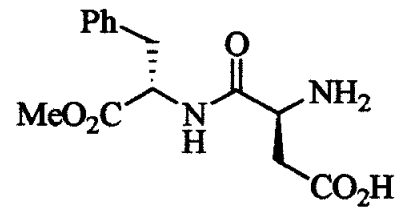
Dextrorfano



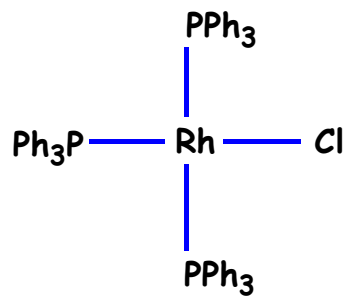
L-DOPA



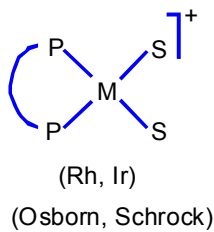
Dopamina



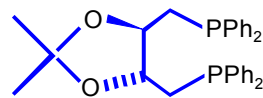
Aspartamo



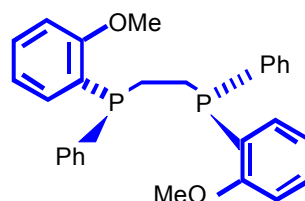
### Catalizador de Wilkinson



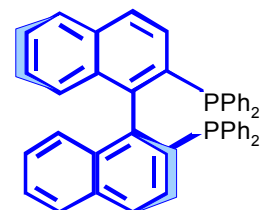
(Rh, Ir)  
(Osborn, Schrock)



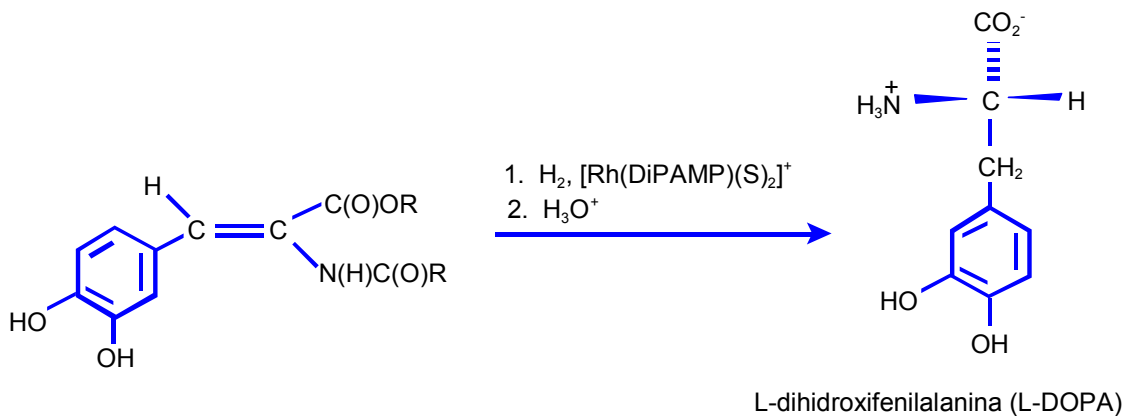
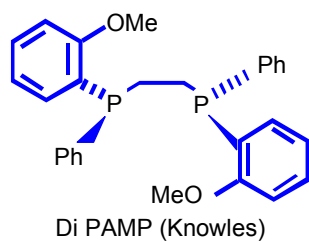
DIOP (Kagan)



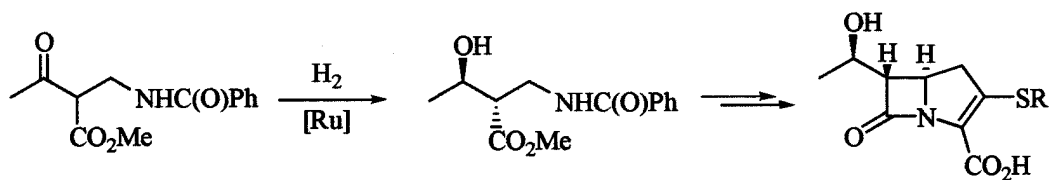
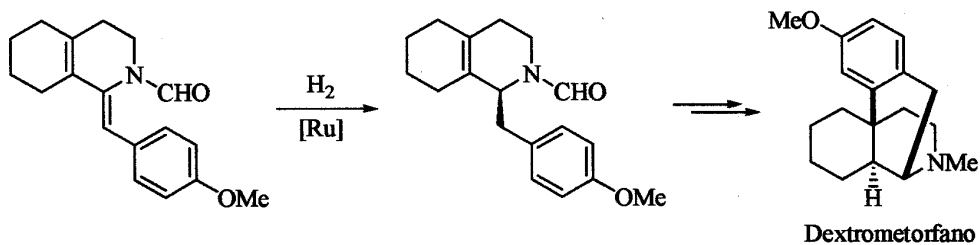
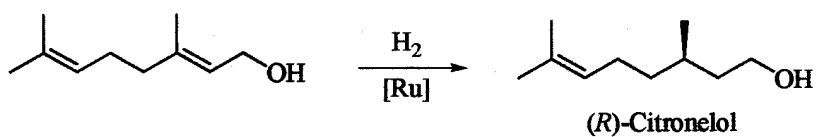
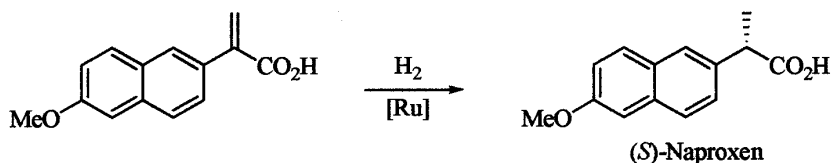
Di PAMP (Knowles)



BINAP (Noyori)

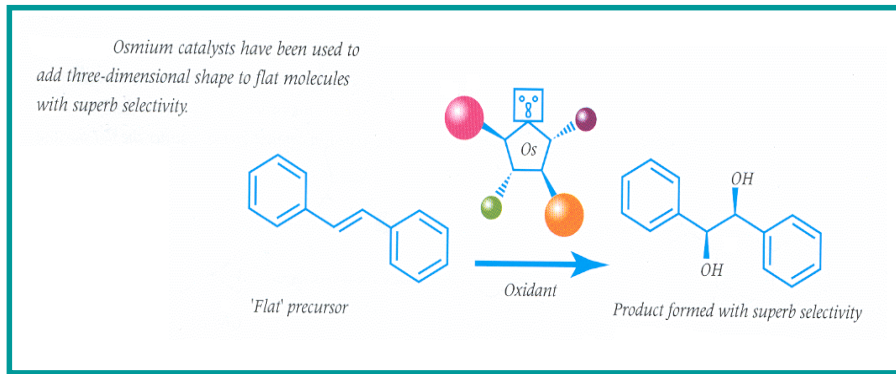
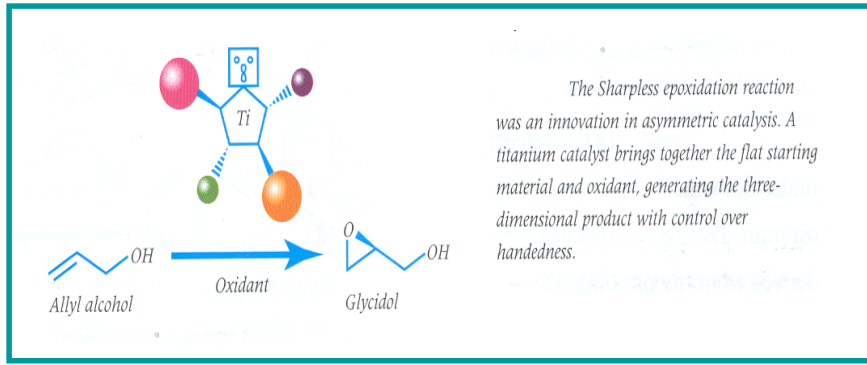


**W. Knowles, Monsanto (1974)**



**R. Noyori**



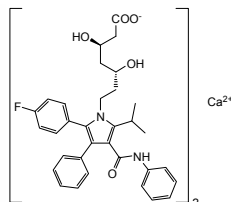


W. Knowles, R. Noyori, B. Sharpless, Premio Nobel de Química, 2001.

## Importancia de los Productos Quirales en la Industria Farmacéutica



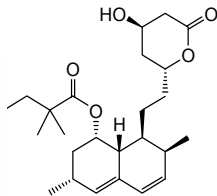
#1  
**LIPITOR**  
Treats: High cholesterol  
Pfizer  
\$8.4 billion  
+8%  
Source: IMS Health, a healthcare information company, Twelve Months Ending December 2005



Atorvastatina



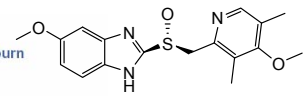
#2  
**ZOCOR**  
Treats: High cholesterol  
Merck  
\$4.4 billion (2005 sales)  
-5%  
Source: IMS Health, a healthcare information company, Twelve Months Ending December 2005



Simvastatina



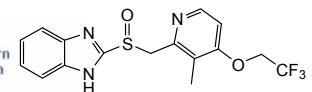
#3  
**NEXIUM**  
Treats: Heartburn  
AstraZeneca  
\$4.4 billion  
+15%  
Source: IMS Health, a healthcare information company, Twelve Months Ending December 2005



Esomeprazol



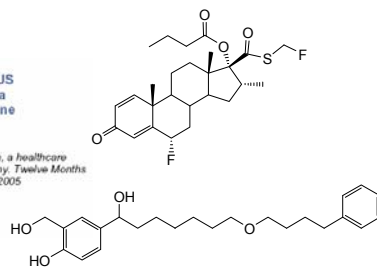
#4  
**PREVACID**  
Treats: Heartburn  
Abbott & Takeda  
\$3.8 billion  
-2%  
Source: IMS Health, a healthcare information company, Twelve Months Ending December 2005



Lanzoprazol

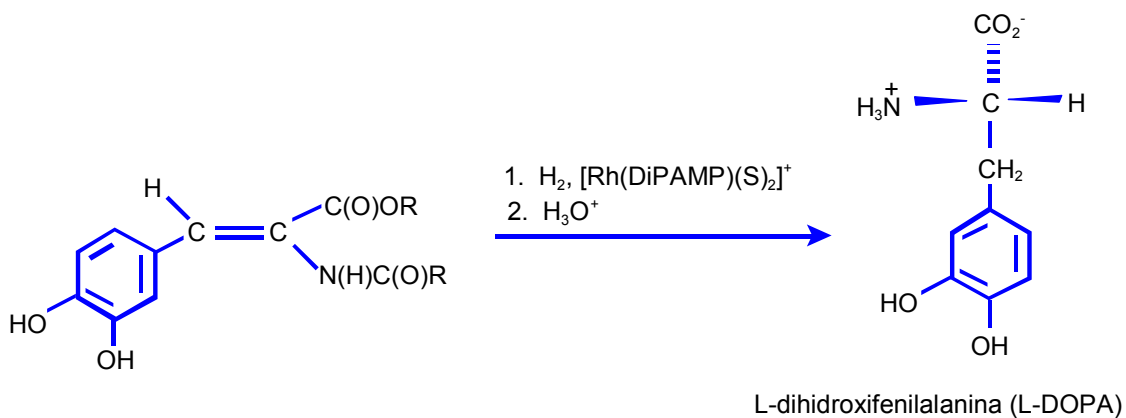
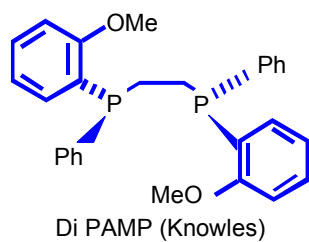


#5  
**ADVAIR DISKUS**  
Treats: Asthma  
GlaxoSmithKline  
\$3.6 billion  
+22%  
Source: IMS Health, a healthcare information company, Twelve Months Ending December 2005

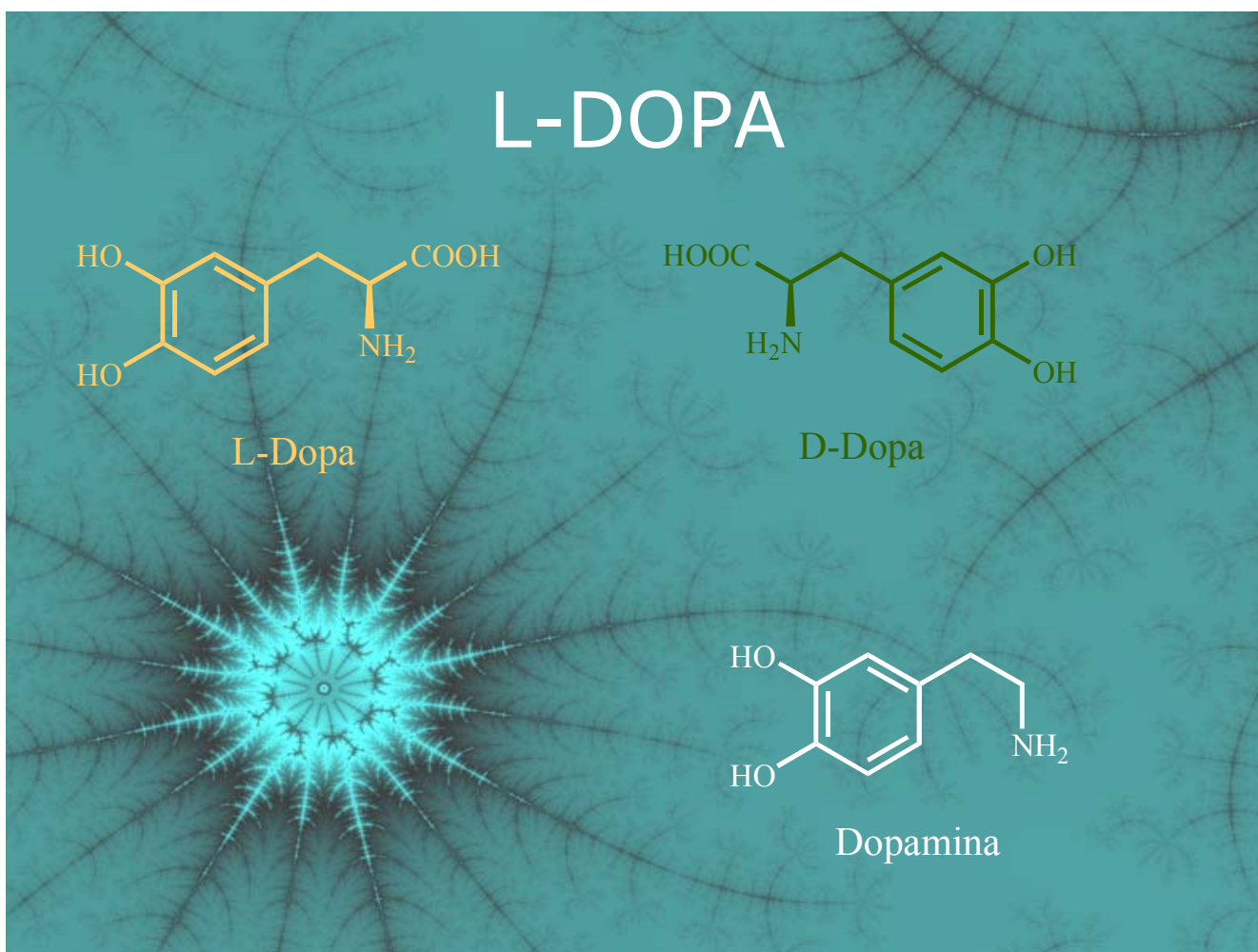


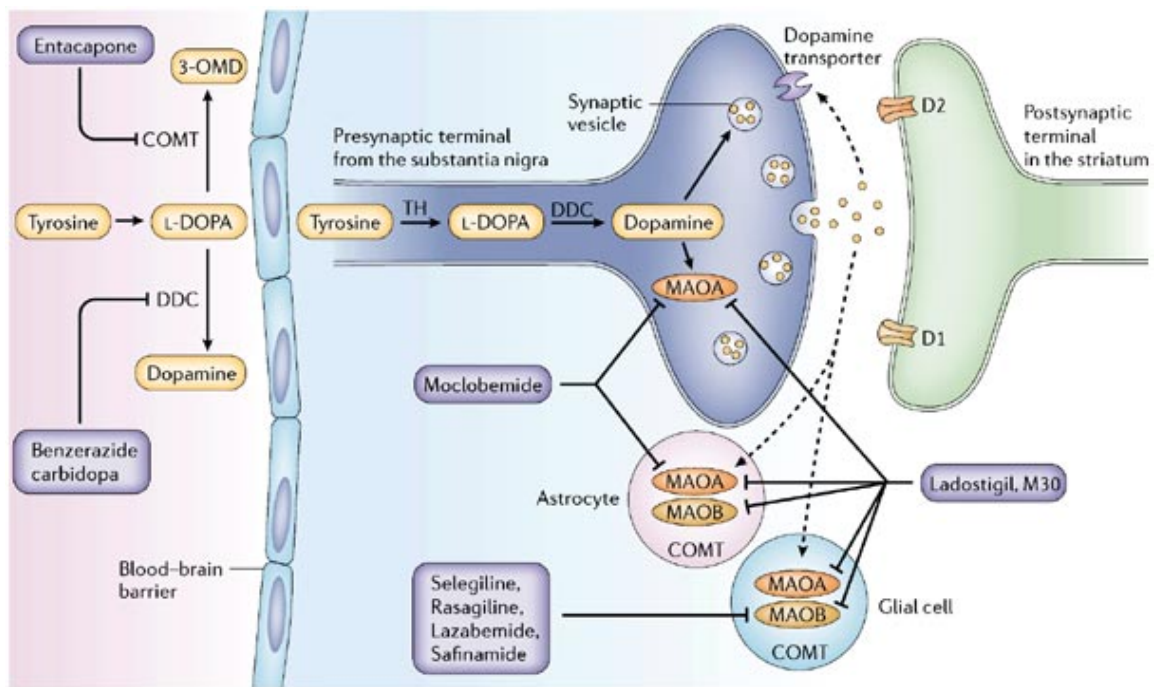
Fluticasona+Salmeterol

Total ventas productos 1-5:  
17.600 millones €

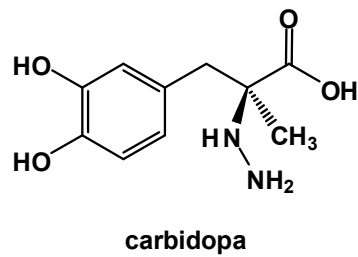
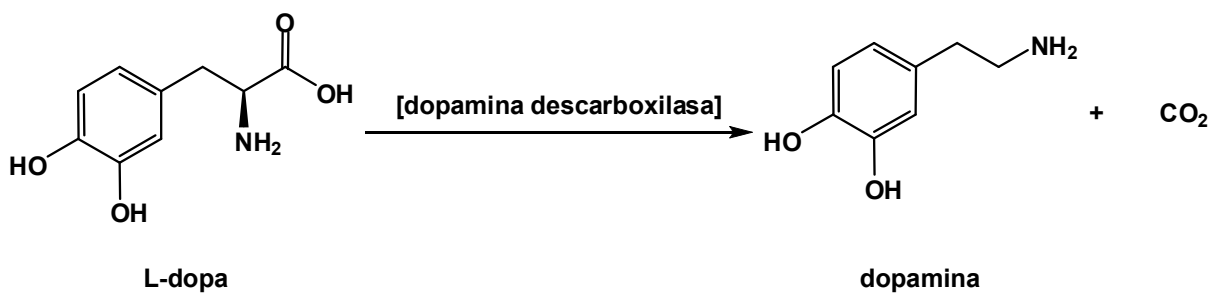


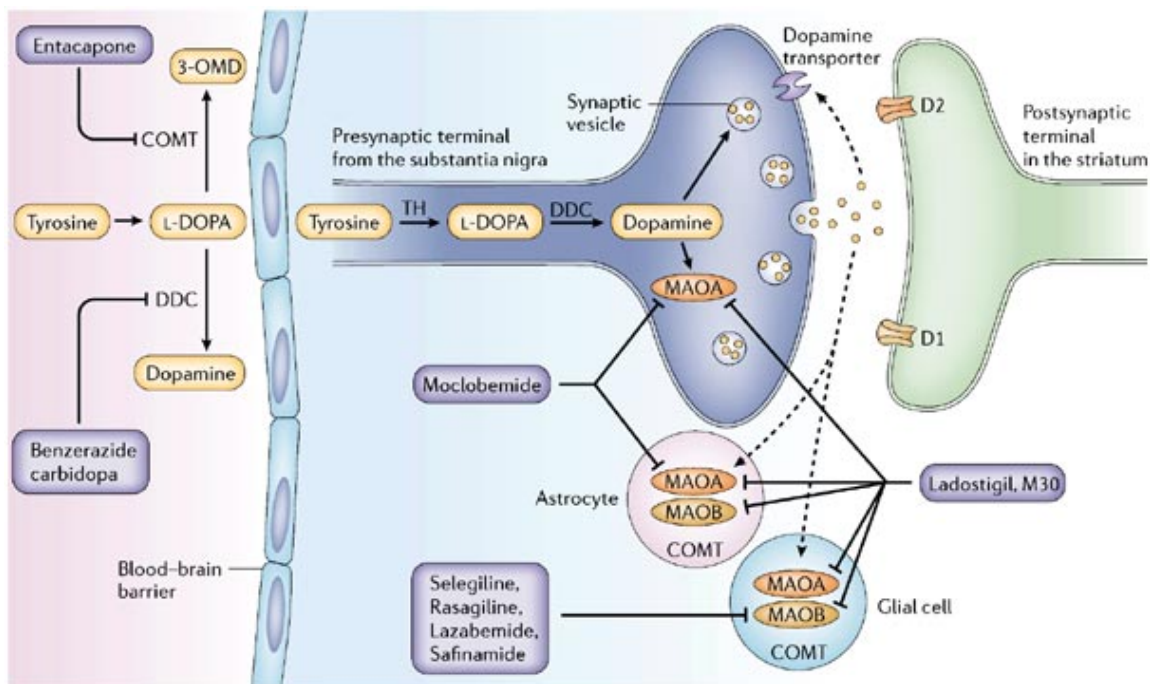
**W. Knowles, Monsanto (1974)**





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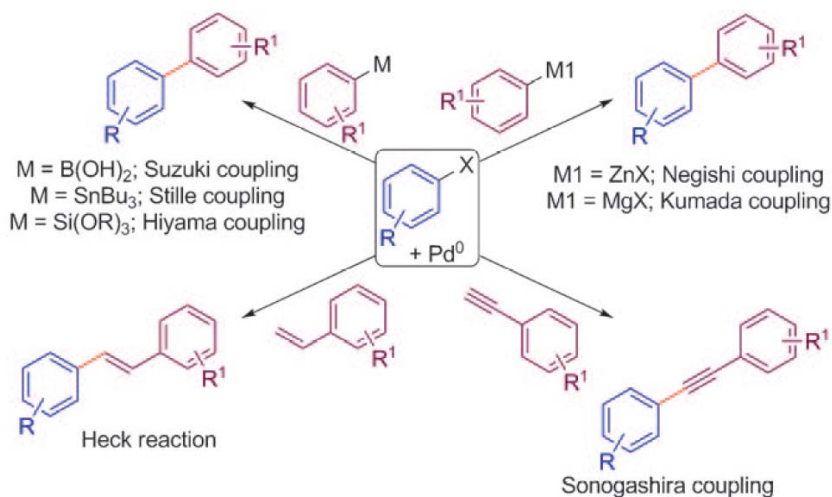
Richard F. Heck



Ei-ichi Negishi



Akira Suzuki



**VERSATILITY** Heck, Negishi, and Suzuki couplings have been used to make various fine chemicals.

