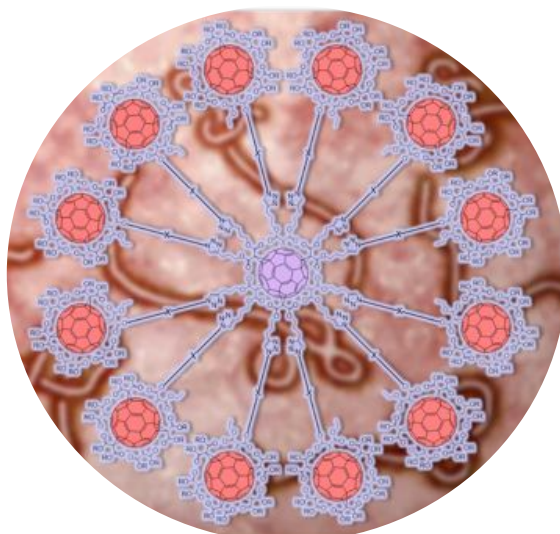
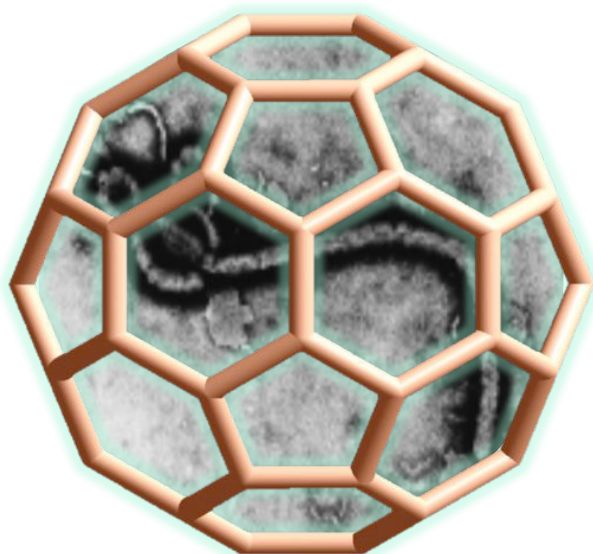
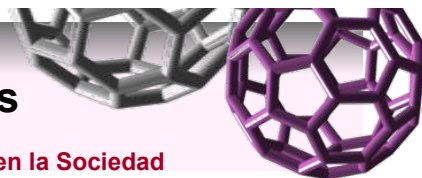




Fullerenos para aplicaciones biológicas

V Curso de Divulgación: Los Avances de la Química y su Impacto en la Sociedad

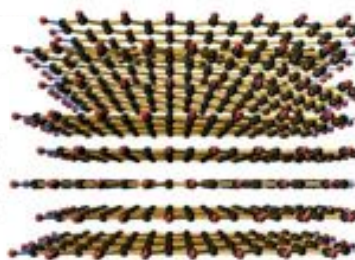


Nazario Martín

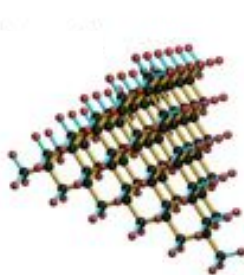
*Departamento de Química Orgánica. Facultad de
Ciencias Químicas
Universidad Complutense de Madrid/IMDEA-
Nanociencia*



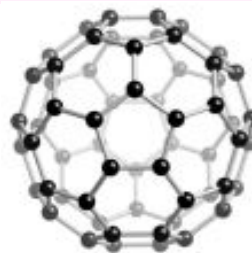
The fascinating forms of carbon...



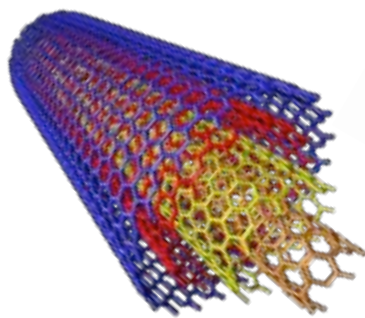
Graphite



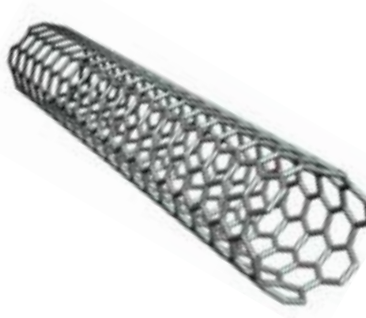
Diamond



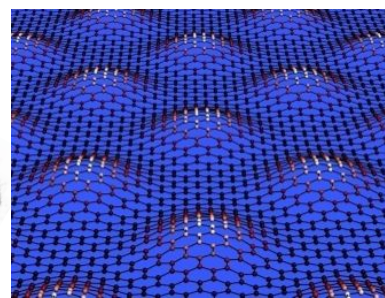
Fullerenes (1985)
Kroto, Smalley, Curl, 1996



Multiwall carbon
nanotubes (1991)



Singlewall carbon
nanotubes (1993)



Graphene (2004)
Geim, Novoselov, 2010

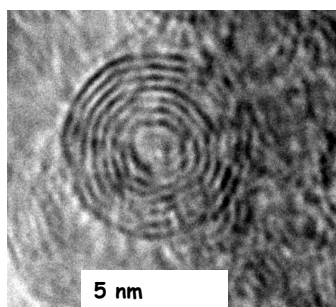
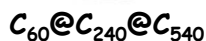
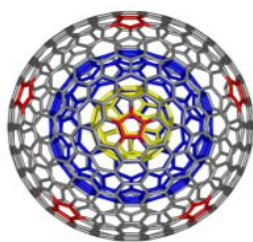
J. Mater. Chem. **2008**, *18*, 1415-1592
Special issue on Carbon Nanostructures



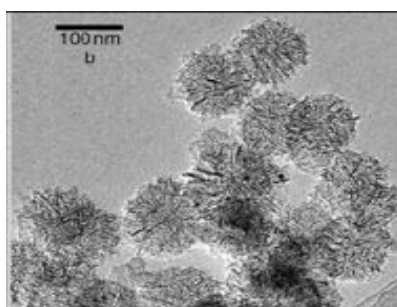
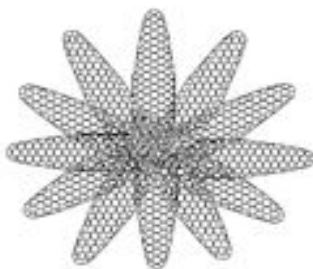
The fascinating forms of carbon...



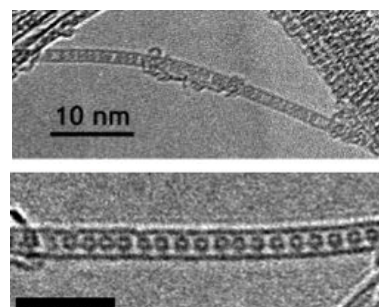
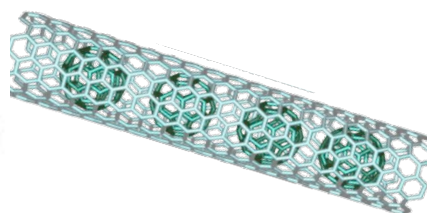
Carbon nanoonions



Nanohorns



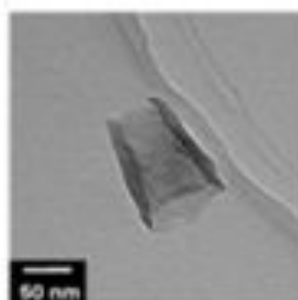
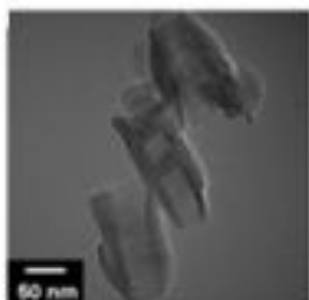
Peapods



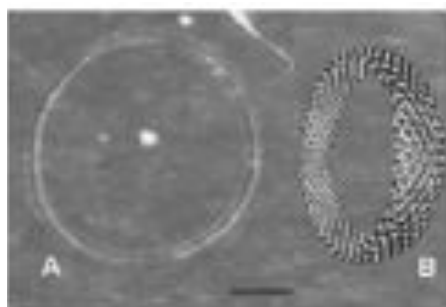
J. Mater. Chem. **2008**, *18*, 1415-1592
Special issue on Carbon Nanostructures



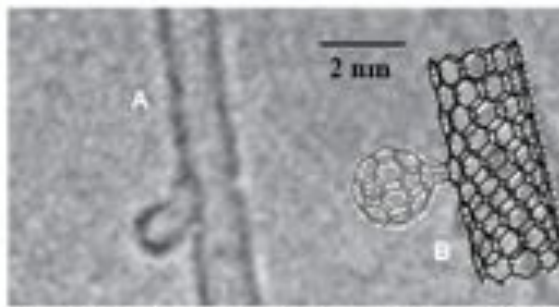
The fascinating forms of carbon...



Nanocups



Nanotorus

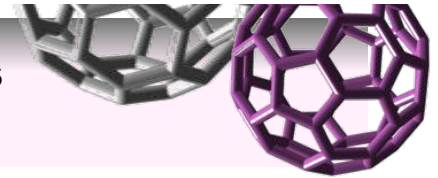


Nanobuds

J. Mater. Chem. **2008**, *18*, 1415-1592
Special issue on Carbon Nanostructures

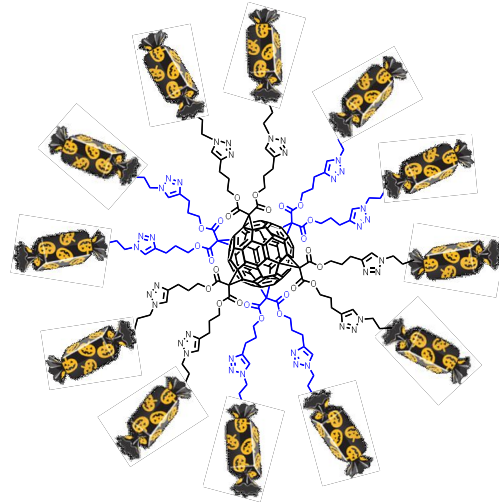
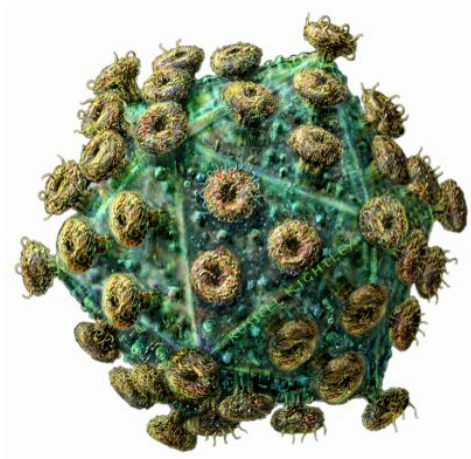


Fullerenos para aplicaciones biológicas

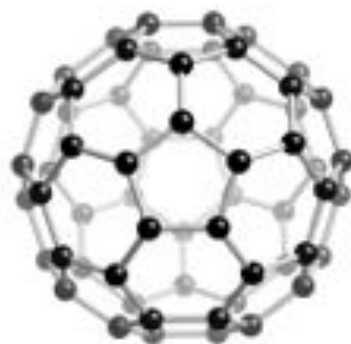
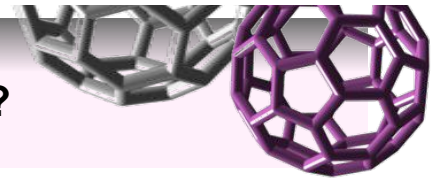


Content:

- Introduction (short) to the use of Fullerenes for biological applications
- Carbohydrate-protein interactions
- Synthesis of [60]Fullerene derivatives as “multivalent” Scaffold
- Glycofullerenes as new and potent inhibitors of viral Infection
- Other carbon nanoforms (SWCNT and SWCNH) as “multivalent” scaffolds



Biological Applications of Fullerenes?

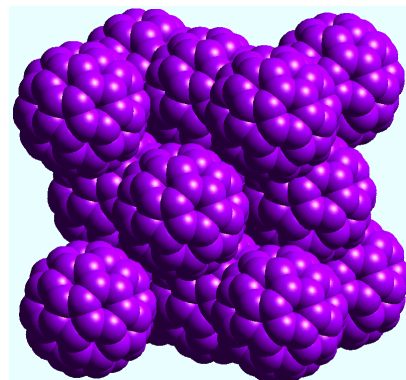
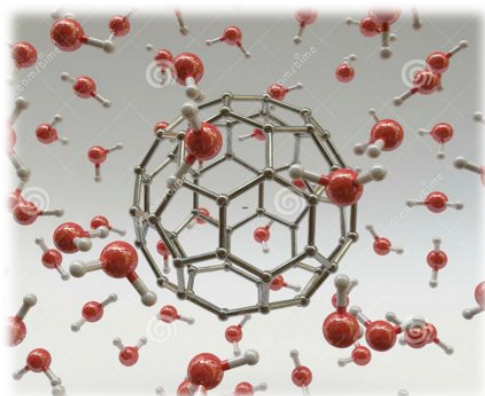
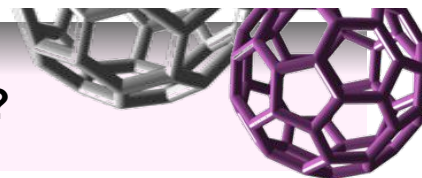


C_{60} in toluene

- Fullerene C_{60} is a **toxic molecule** which is insoluble in water!!!



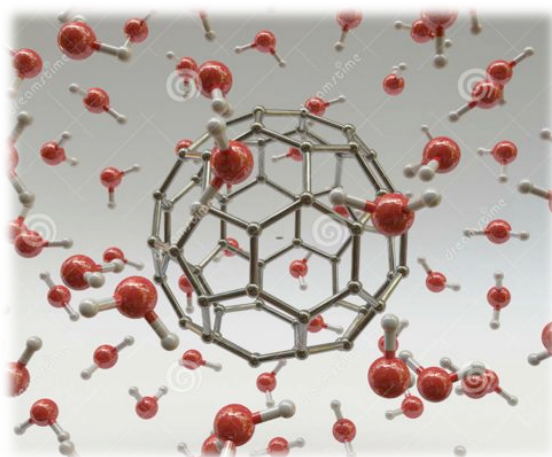
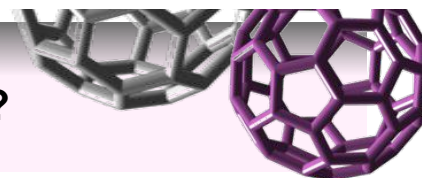
Biological Applications of Fullerenes?



- Fullerenes in contact with water form toxic colloidal aggregates (nC_{60}).
- Clusters of C_{60} can cause oxidative damage to lipids in the brains of fish.
- C_{60} is highly hydrophobic and redox active, therefore, it can potentially cause oxidative damage. Water dispersed C_{60} provokes oxidative stress in the brain of largemouth bass and depletion of glutathione in the gills of fish.
- This *in vivo* study, showing the adverse effect of C_{60} in aquatic species, may predict potential effects in humans.



Biological Applications of Fullerenes?

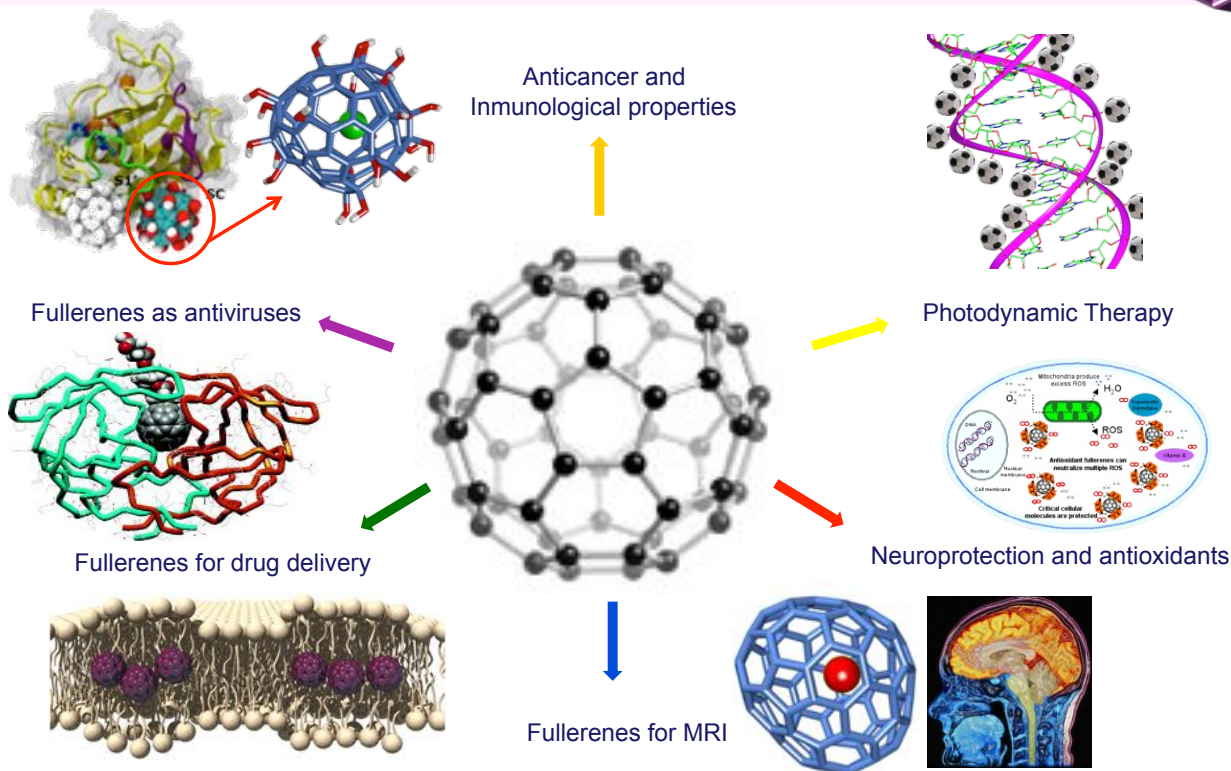
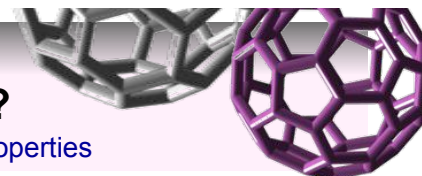


- Fullerene C_{60} is a **toxic molecule** which is insoluble in water!!!
- Therefore, different strategies have been explored to **render fullerenes biocompatible**.
- **Solubilization** and **chemical functionalization** are the starting point to integrate this material into living systems (aminoacids, sugars, cyclodextrins, etc)

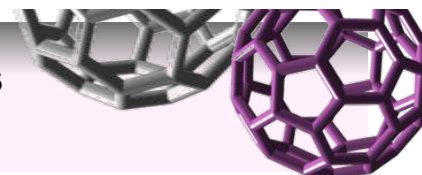


Biological Applications of Fullerenes?

Suitably functionalized, fullerenes show interesting biological properties

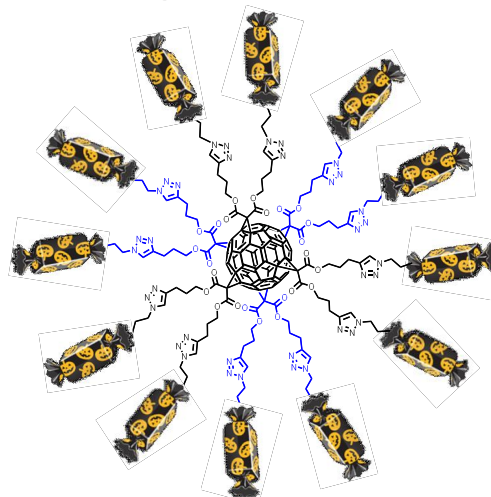
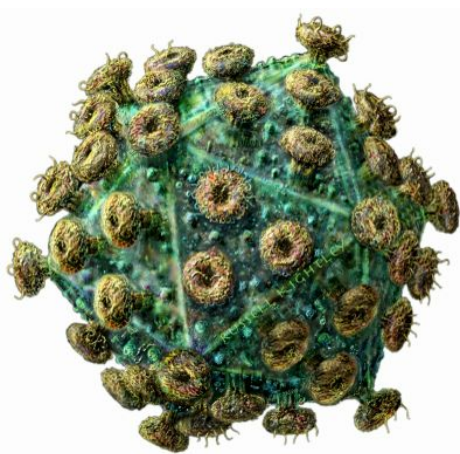


Fullerenos para aplicaciones biológicas



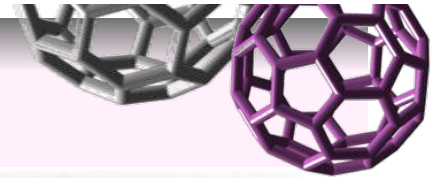
Content:

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- Other carbon nanoforms (SWCNT and SWCNH) as “multivalent” scaffolds



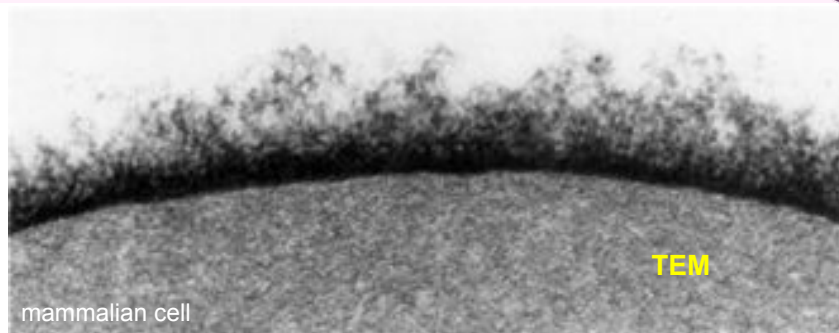


Carbohydrate Interactions



Glycocalyx

- Protection
- Social Behaviour

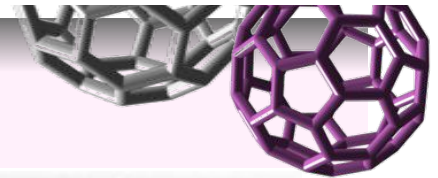


Functions of glycocalyx:

- **Protection:** Cushions the plasma membrane and protects it from chemical injury.
- **Immunity to infection:** Enables the immune system to recognize and selectively attack foreign organisms.
- **Defense against cancer:** Changes in the glycocalyx of cancerous cells enable the immune system to recognize and destroy them.
- **Transplant compatibility:** Forms the basis for compatibility of blood transfusions, and organ transplants.
- **Adhesion:** Binds cells together so that tissues do not fall apart
- **Inflammation regulation:** Glycocalyx coating on endothelial walls in blood vessels prevents leukocytes from rolling/binding in healthy states
- **Fertilization:** Enables sperm to recognize and bind to eggs
- **Embryonic development:** Guides embryonic cells to their destinations in the body.

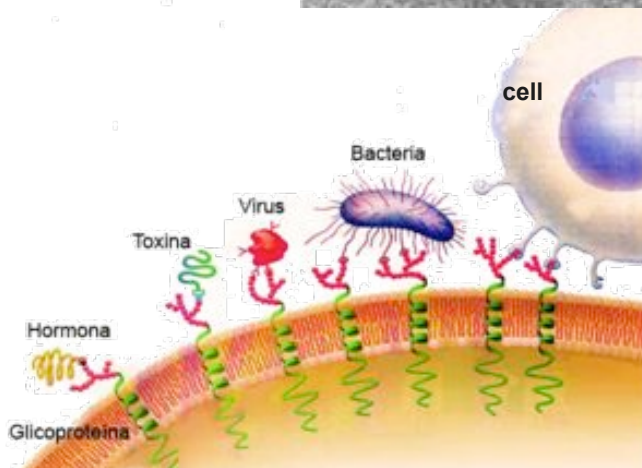
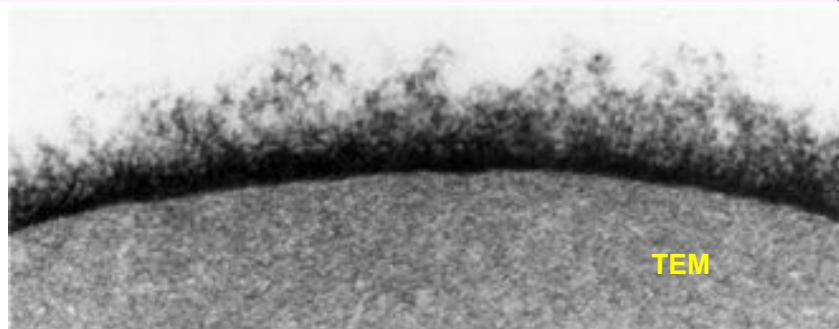


Carbohydrate Interactions



Glycocalyx

- Protection
- Social Behaviour



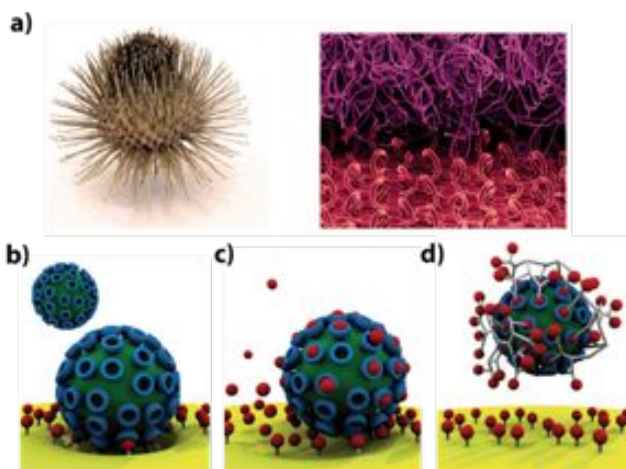
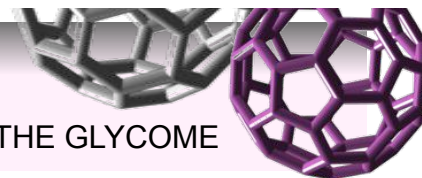
Carbohydrate-protein Interactions

- Highly selective
- Calcium dependent
- Low affinity
- Multivalent interaction



Multivalence

MULTIVALENT SYSTEMS AND INTERACTION WITH THE GLYCOME



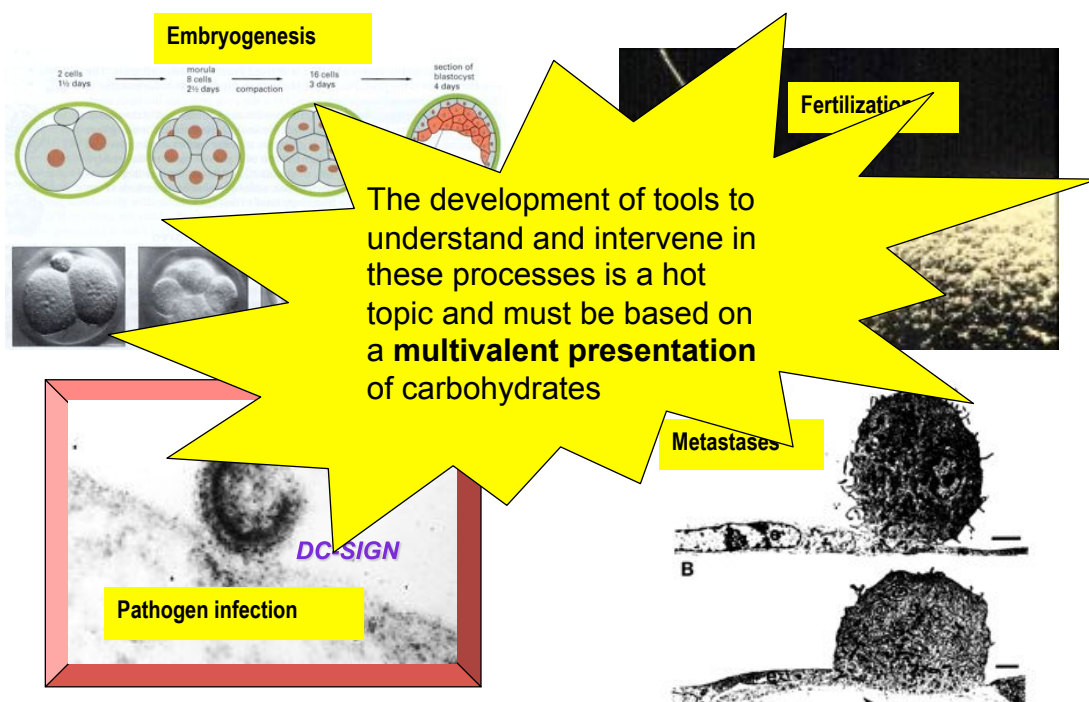
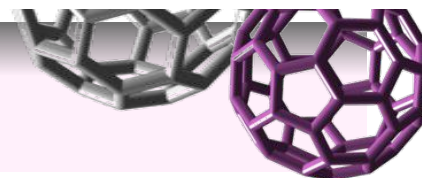
- a) Bardane fruits and Velcro™
- b) A virus connecting to many cellular receptors
- c) Linkage in the presence of a monovalent inhibitor (farmacological classical approach)
- d) Inhibiting the linkage and infection by using a multivalent inhibitor

R. Haag *et al.* *Angew. Chem. Int. Ed.* **2012**, 51, 10472.



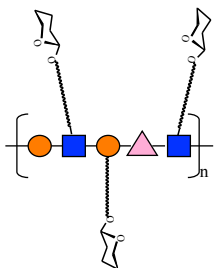
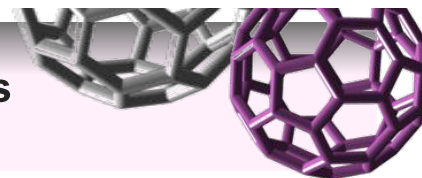
Biological examples

(natural and pathogenic processes)

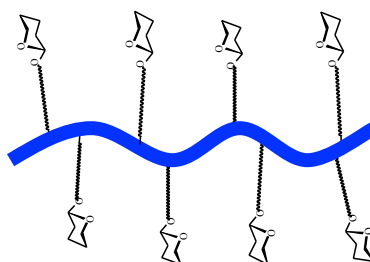




Carbohydrate Multivalent Architectures (synthetic scaffolds)



Glycopetides and
Glycoproteins



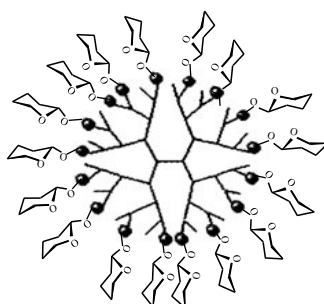
Glycopolymers



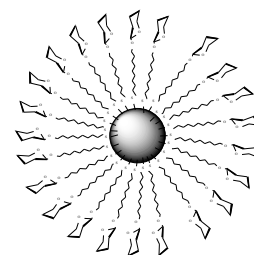
Glycocyclodextrines
and glycolixarenes



Glycoliposomes



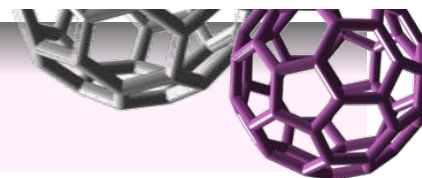
Glycodendrimers



Glyconanoparticles

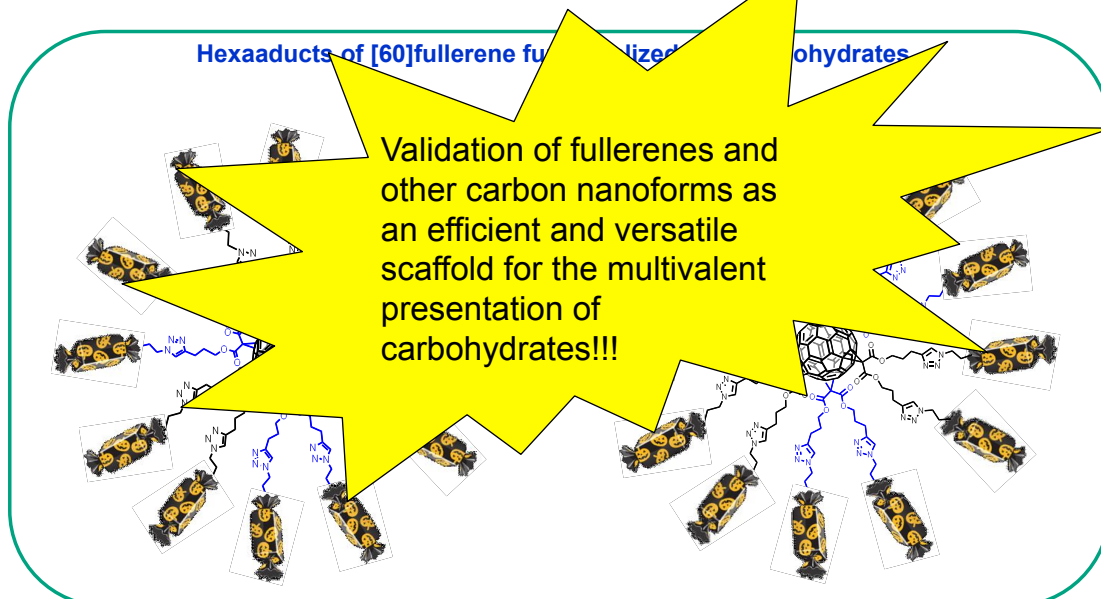


Goals



SYNTHESIS OF NEW WATER SOLUBLE C₆₀ DERIVATIVES

(biocompatibility and toxicity effects. Globular shape and size control)



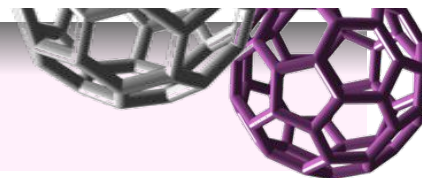
Synthetic approach based on Bingel reaction and further 1,3-dipolar cycloadditions between alkynes and azides catalyzed by Cu(I) (CuAAC).



Synthesis

CHEMICAL MODIFICATION OF FULLERENES:

Toward water soluble fullerenes



Optimizing the CuAAC reaction



Click
Chemistry

Reaction conditions:

Catalyst: $\text{CuBr} \cdot \text{S}(\text{CH}_3)_2 / \text{Cu}^0$

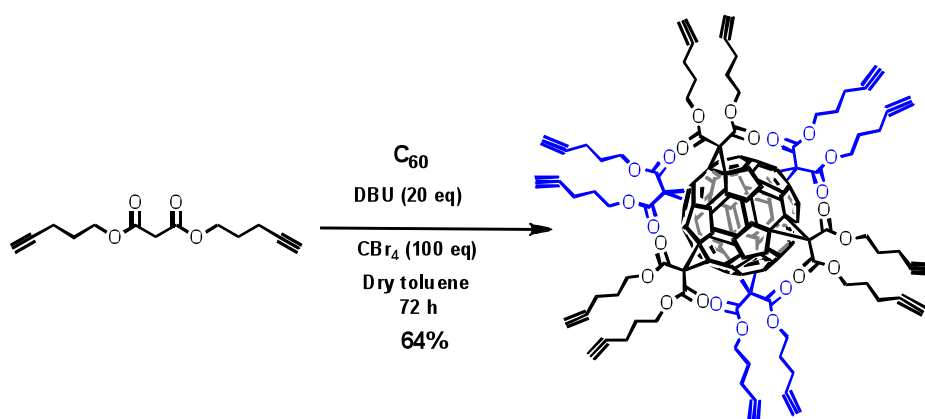
Solvent: DMSO

Reductant: Sodium ascorbate



Synthesis

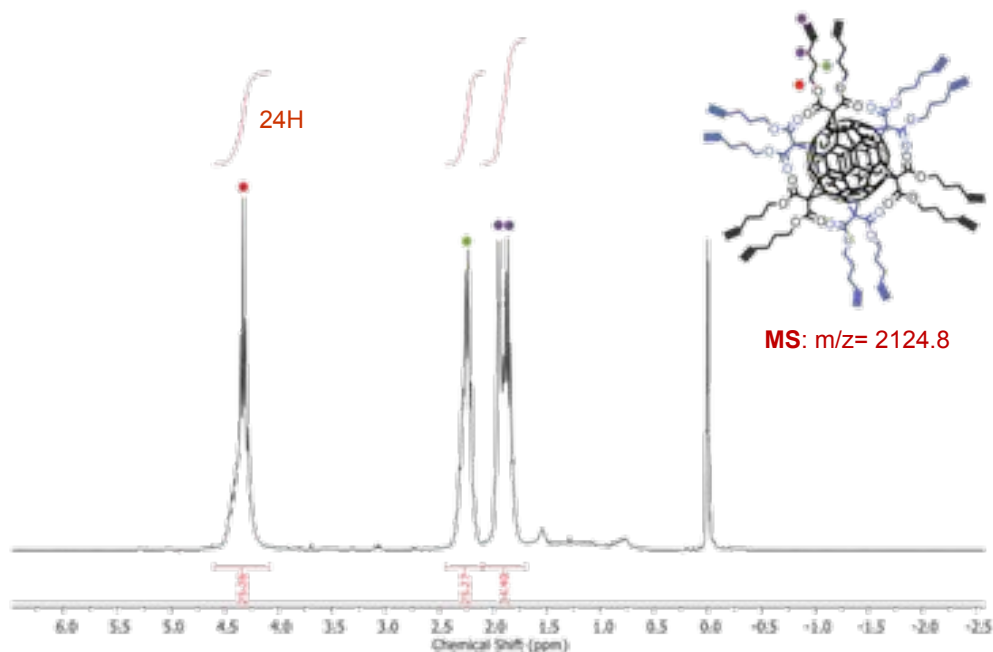
SYNTHESIS AND CHARACTERIZATION OF C_{60} HEXAADUCTS





Synthesis

SYNTHESIS AND CHARACTERIZATION OF C₆₀ HEXAADUCTS



¹H-NMR (700 MHz, CDCl₃, 298 K) of alkyne-substituted hexaadduct showing its 84 protons

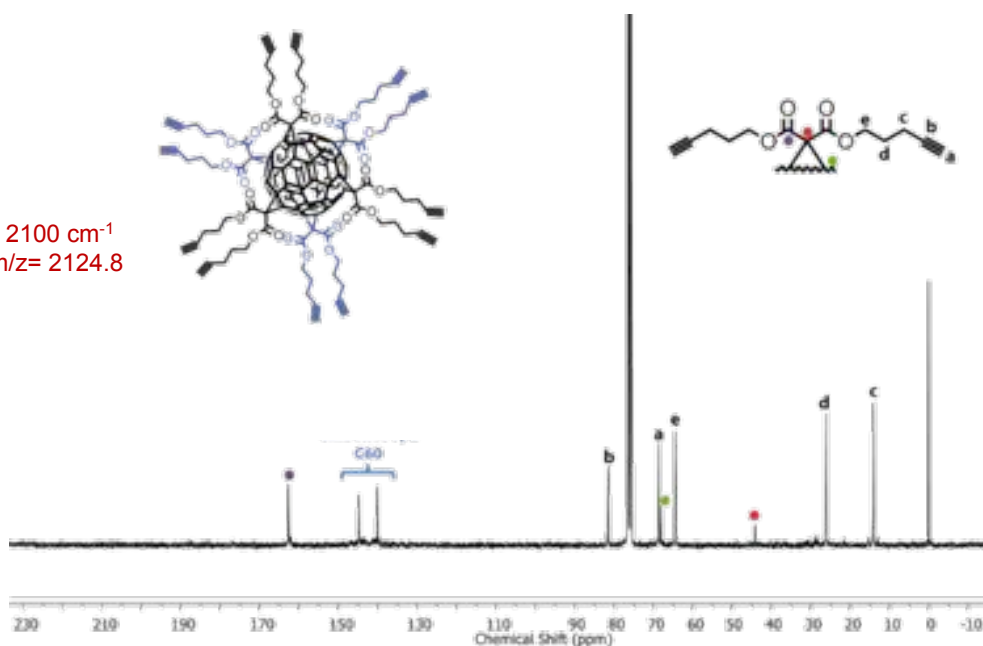


Synthesis

SYNTHESIS AND CHARACTERIZATION OF C₆₀ HEXAADUCTS



FTIR: 2100 cm⁻¹
MS: m/z= 2124.8

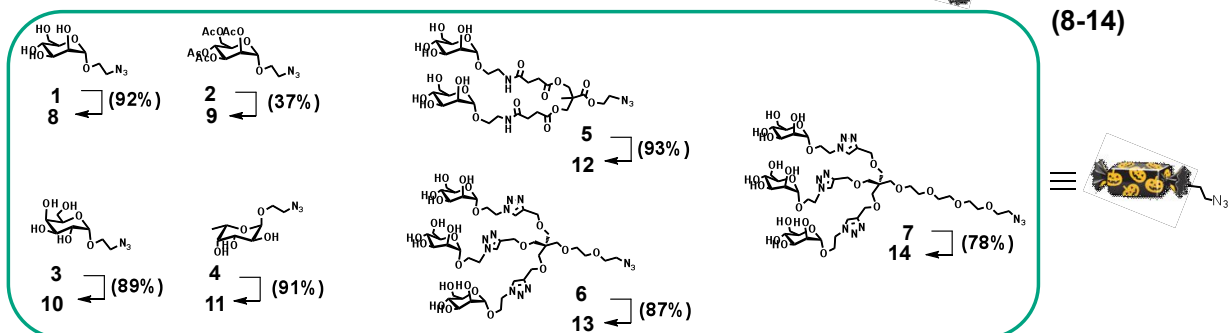
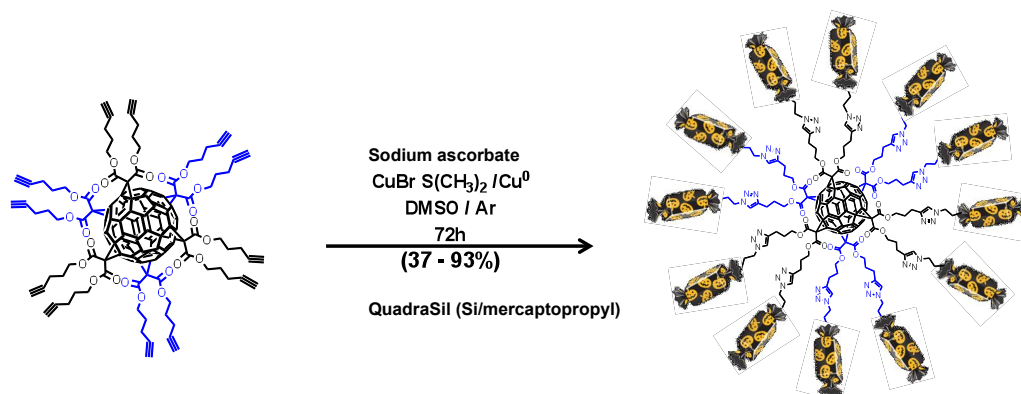


¹³C-NMR (175 MHz, CDCl₃, 298 K) of alkyne substituted hexaadduct (**only 2 sp² carbons of fullerene!!!**)



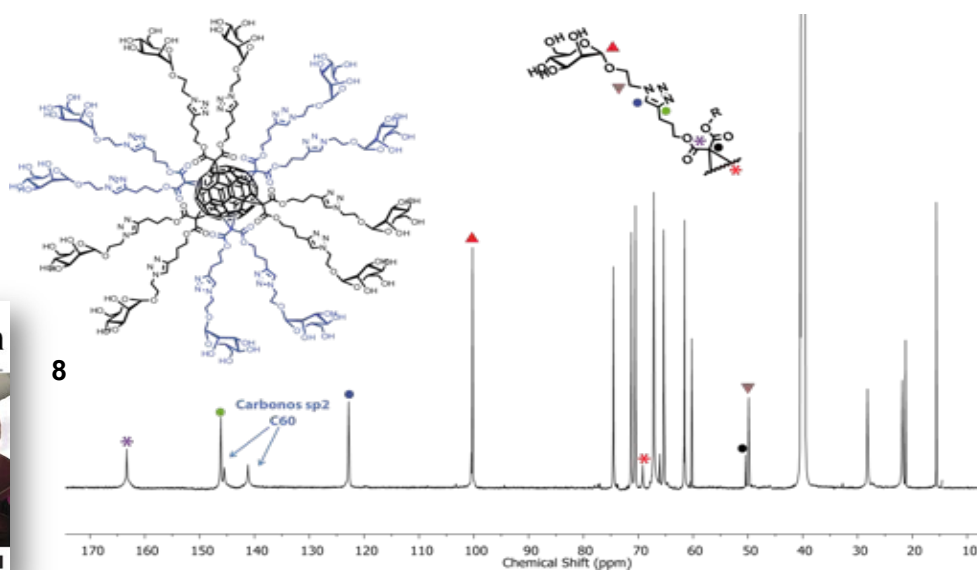
Synthesis

SYNTHESIS AND CHARACTERIZATION OF C₆₀ HEXAADUCTS



Synthesis

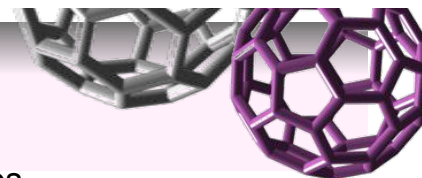
SYNTHESIS AND CHARACTERIZATION OF C₆₀ HEXAADUCTS



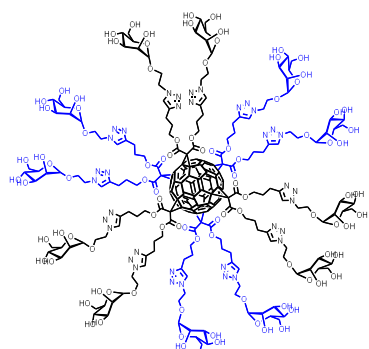
¹³C-NMR spectrum (175 MHz, DMSO-d₆, 298 K) of compound 8



Properties



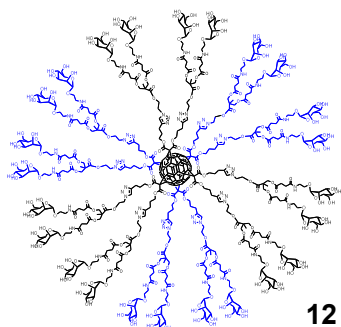
Interaction between lectines and glycofullerenes



8



- Excellent multivalent ligand
- Validation of fullerenes as efficient platform for the multivalent presentation of carbohydrates



12

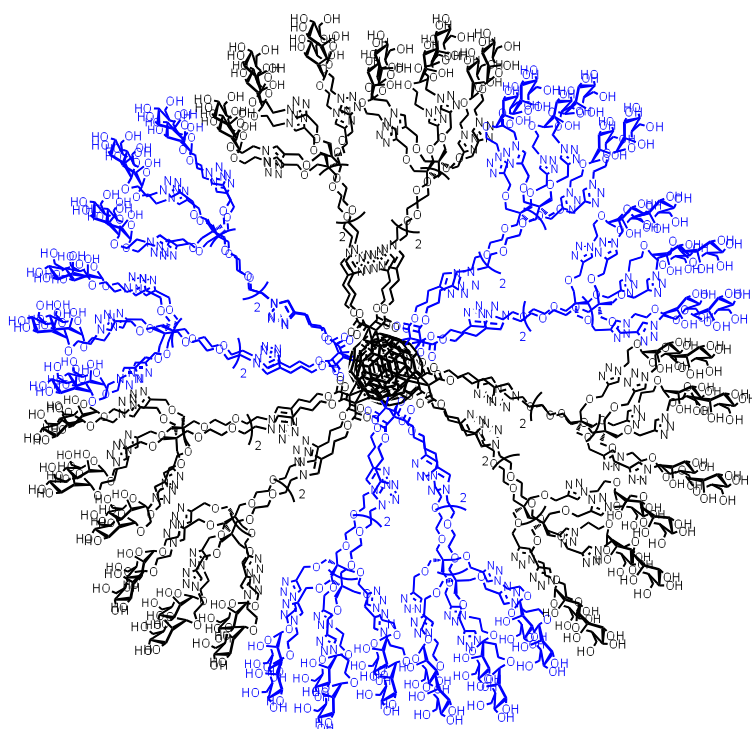
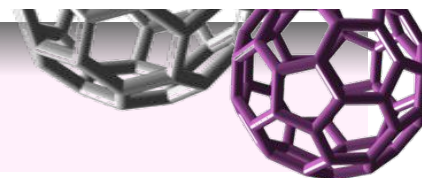


- Increased valence in carbohydrates, but poorer interaction with Con A
- Possible steric congestion and self-association phenomena

Chem. Eur. J. **2011**, *17*, 766.



Diferent Molecular Structures



**36 CARBOHYDRATES
PER
MOLECULE!!**

**We increase the number of
carbohydrates per molecule as
well as the spacer length...**

Biomacromolecules **2013**, *14*, 431.